

# An alternative approach to partial geochemical extractions:

## *The Pink Leach*

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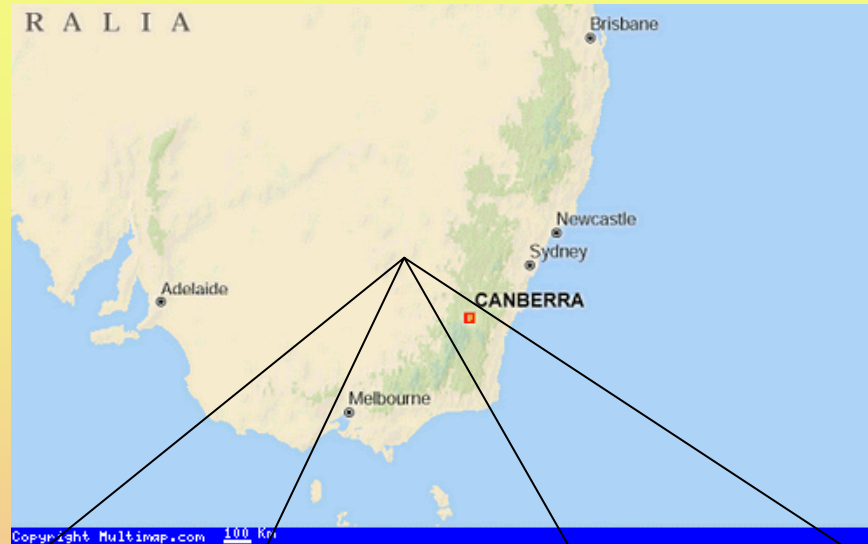
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# Location – Mandamah

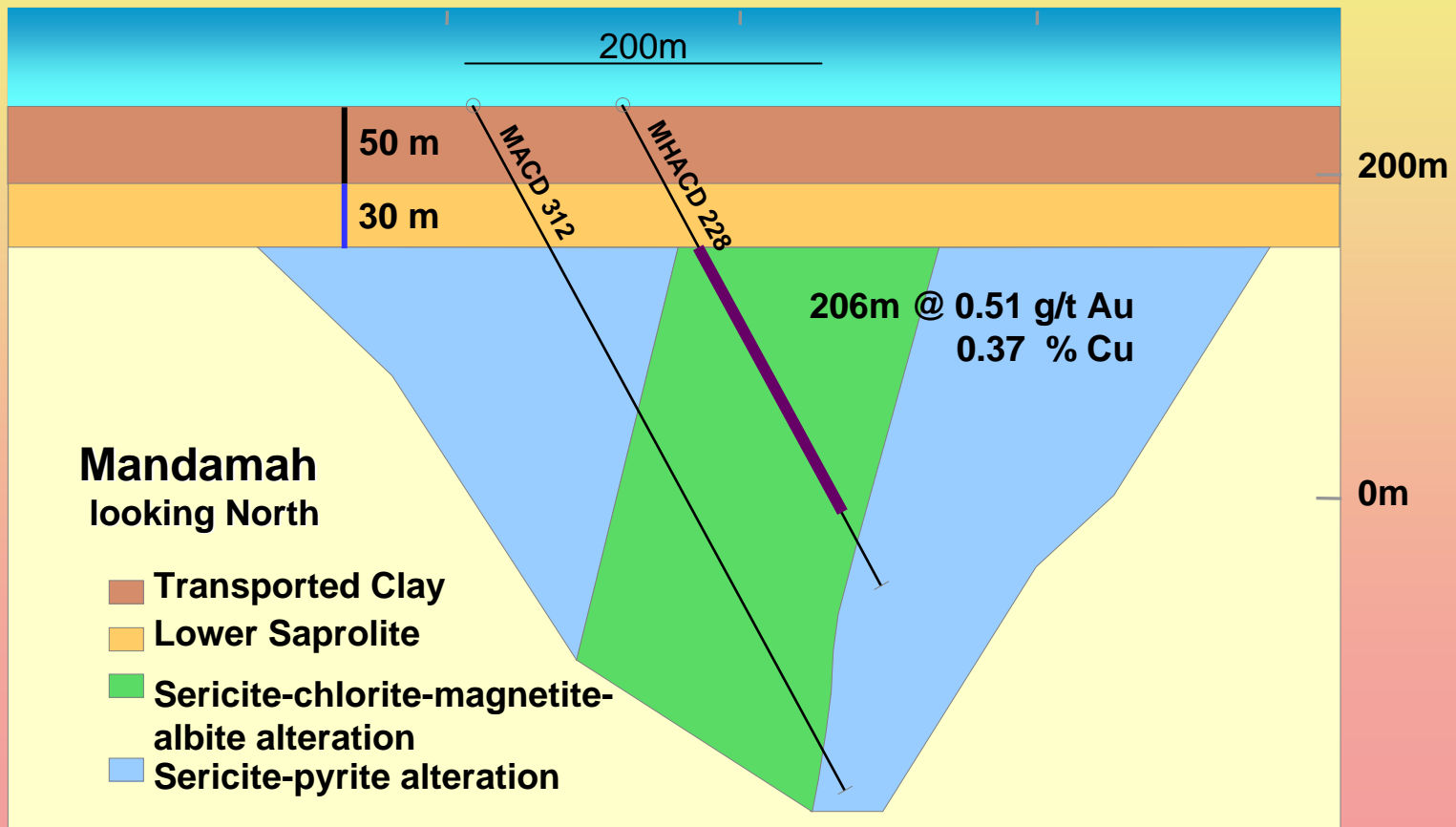


# Mandamah



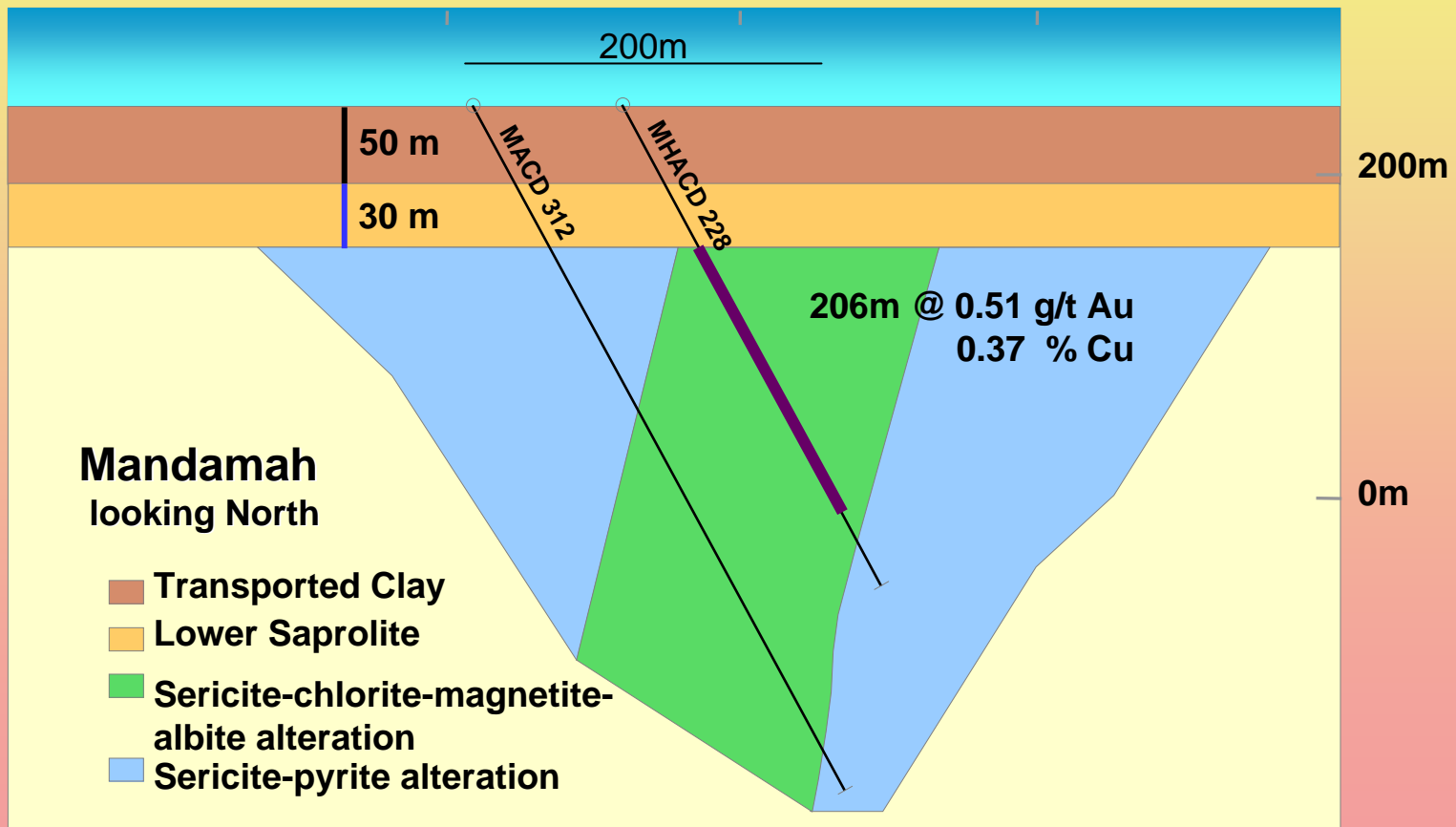
# Mandamah – Geology

- Cu-Au porphyry style mineralisation, possibly E. Silurian.
- Mineralisation hosted in quartz stockworks in coarse plagioclase-phyric quartz diorite.



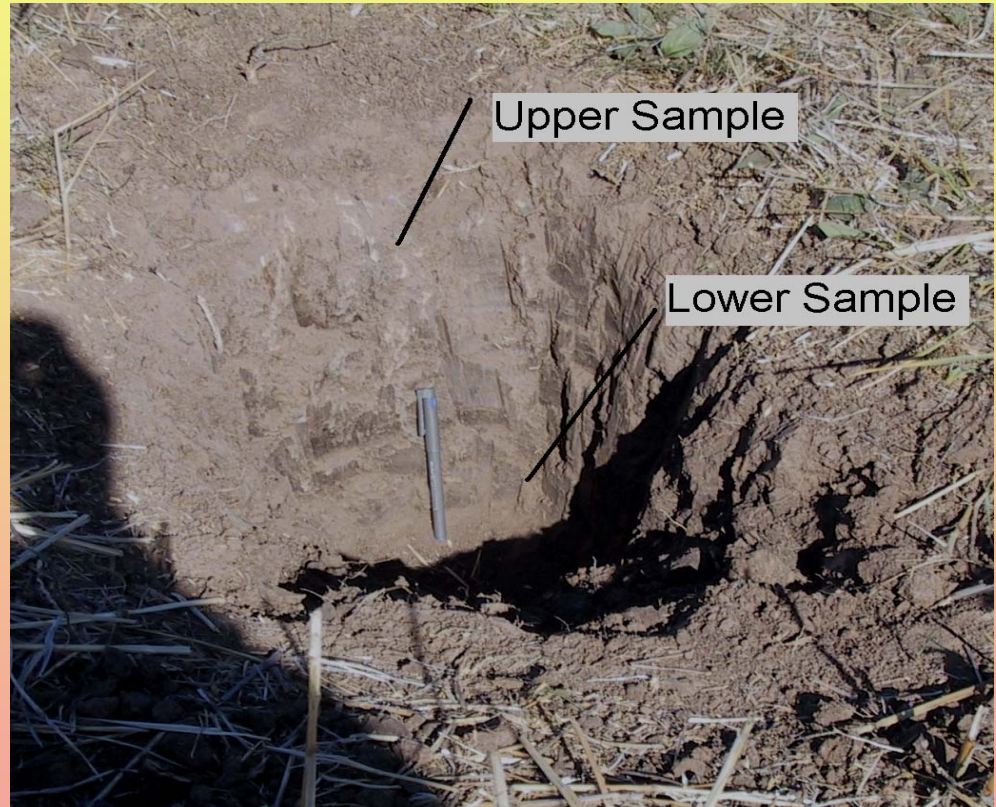
# Mandamah – Geology

- 30m of saprolite developed on the diorite.
- Area covered by ~50m ?post Miocene clay-rich alluvium.
- Supergene Cu Au development at alluv-sap boundary.



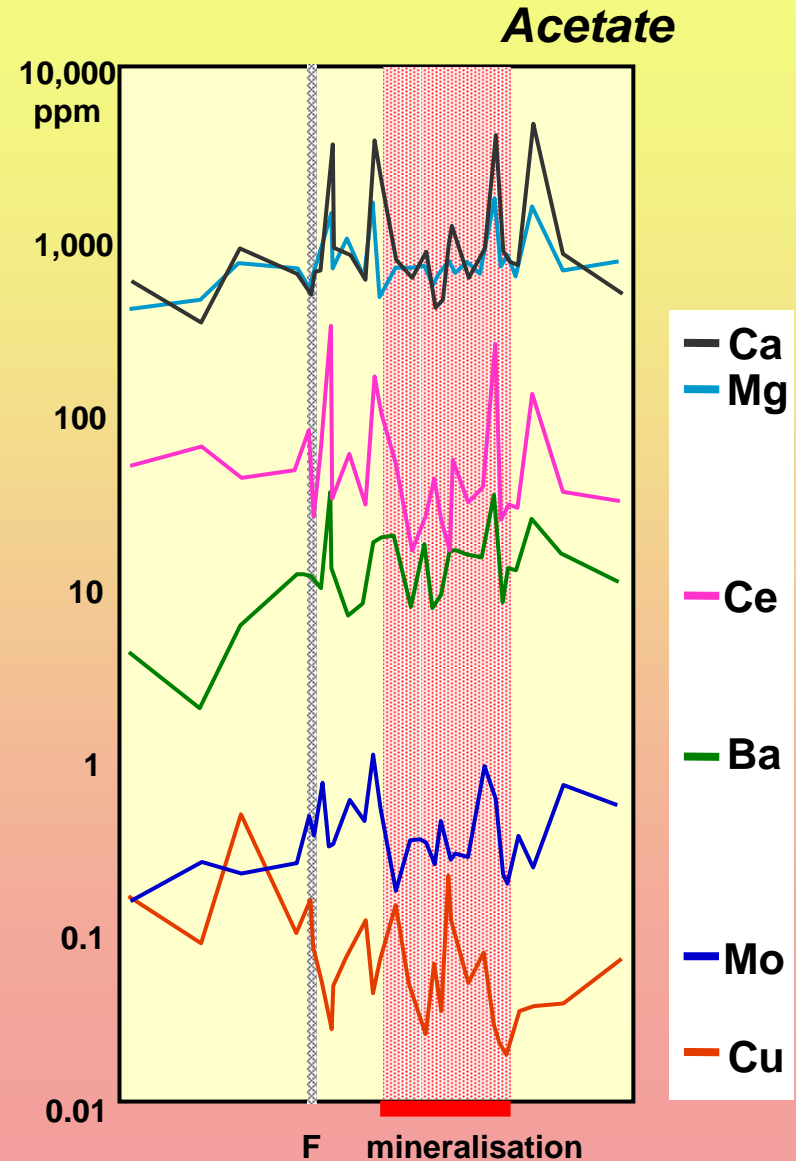
# Mandamah – Samples

- Grey to beige, massive and well sorted alluvium.
- >70% by weight clays and  $-180\mu\text{m}$  quartz fragments.
- Quartz and lithic fragments  $< 1\text{mm}$ .
- Carbonates both as cements and nodules at selected locations.



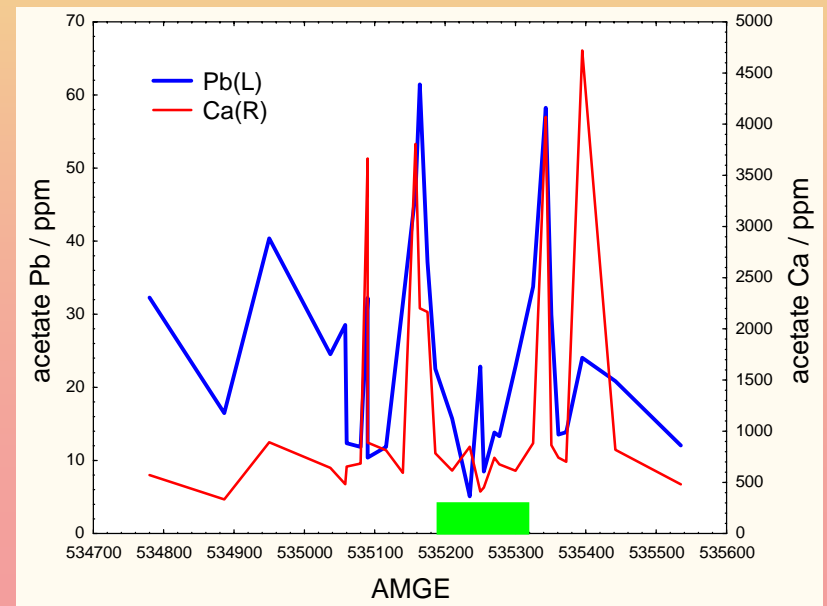
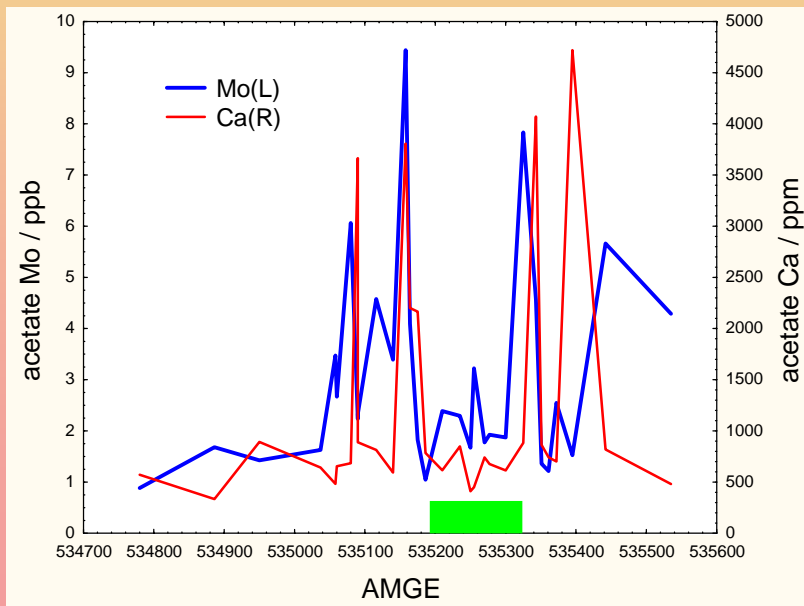
# Mandamah – Orientation Study

- Extensive survey performed with 5 leaches that did not display patterns of ore related metals definitively related to mineralisation.
- Significant variability seen in the traverses, particularly over and around structural features and the periphery of mineralisation.
- Subtle leaches appeared to offer the greatest variability that might be related to mineralisation.



# Mandamah – Orientation Study

- The distribution of many major and trace metals appears to be affected by the distribution of carbonates.
- The carbonate distribution may be affected mineralisation but also may affect the redistribution of trace elements in the sample or extractability in the leach.





# Mandamah – Orientation Study

- It was concluded that we needed a weak leach that would be unaffected by local variations in the soil's capacity to consume acid or base.
- Buffering appeared to not offer the required control in the acetate buffer pH range.
- A leach was required that would be tailored for each sample while not taking an excessive amount of time or significantly affecting the natural background soil chemistry.



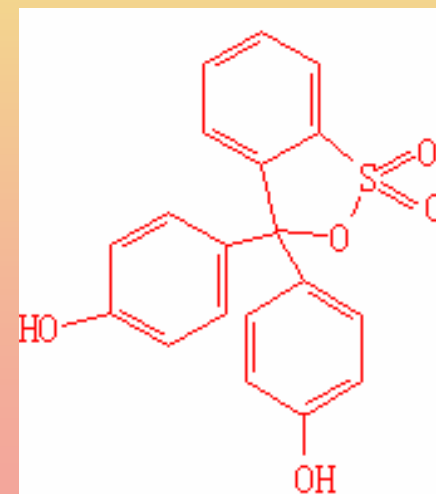
# Pink Leach – Theory

- In regions of uniform cover, local changes in a soil's buffering capacity may be a consequence of an active geochemical system.
- The Pink Leach is a  $\text{Ca}(\text{NO}_3)_2$  based exchange leach with no chelating agents.
- The Pink Leach reacts all samples at the same pH irrespective of the sample alkalinity or acidity.



# Pink Leach – Chemistry

- Concentrated  $\text{Ca}(\text{NO}_3)_2$  solution  $\sim 10\text{g/L}$ .
- Neutral, acidic and alkaline  $\text{Ca}(\text{NO}_3)_2$  solutions are prepared
- Trace concentration of phenol red in each
  - Indicator with a pH range of 6.8 to 8.4
  - Equivalence point pH of 7.3 to 7.8
  - This indicator was chosen as pH slurries for samples distant to mineralisation at Mandamah are within this pH range and most metals are mobile
- This is now referred to as the **PR- $\text{Ca}(\text{NO}_3)_2$**  reagent.



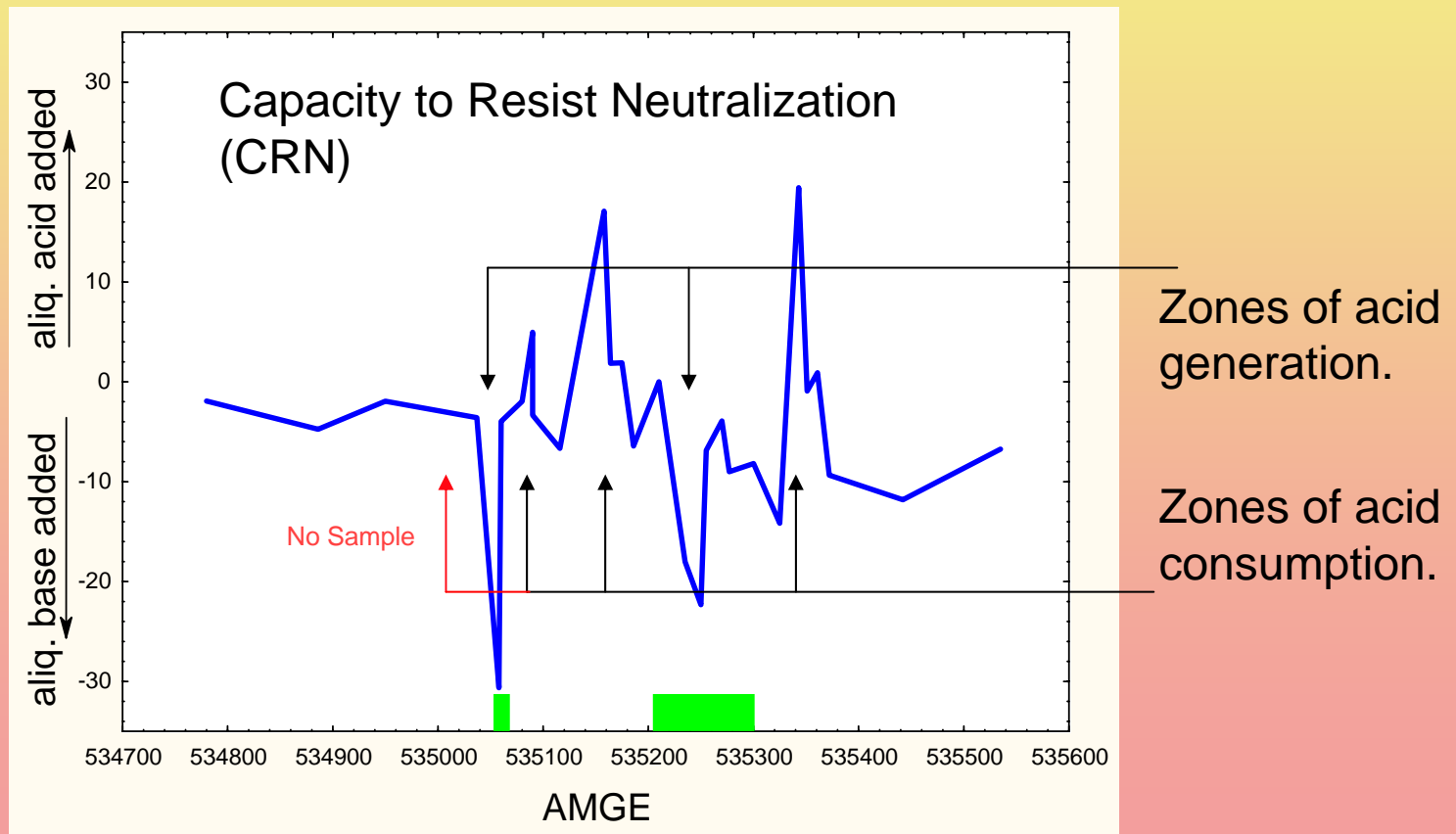
# Pink Leach - Method

- A “standard solution” is made at a known pH in the transition range for the indicator.
- Add 20mL of neutral **PR**-Ca(NO<sub>3</sub>)<sub>2</sub> reagent to 1g of sample.
- The solution pH quickly equilibrates with the sample.
- If the solution is yellow, it is too acidic.
- If the solution is red, it solution is too alkaline.
- Acidic or alkaline **PR**-Ca(NO<sub>3</sub>)<sub>2</sub> reagent is added in small measured quantities until the colour of solution matches the standard solution.
- The leachate is removed when the fine material flocculates, typically within 5 minutes.



# Pink Leach - Results

- Addition of acid or base indicates a capacity to resist change in soil conditions.



# Pink Leach - Results

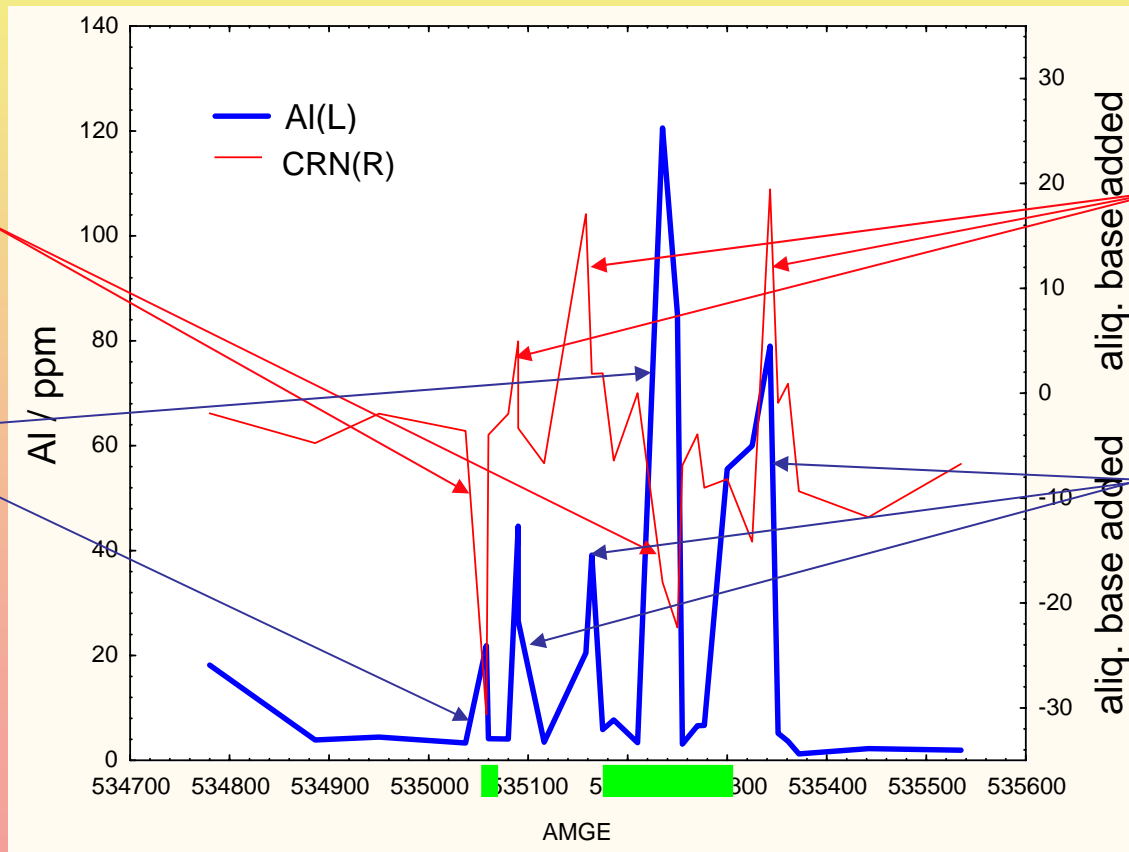
- Aliquot measurement is significantly different to a simple pH measurement on a slurry.
- pH measurements give an indication of the dominant acid-base species in a sample.
- The amount of acid or base required to neutralise a sample is a measure of the soil's capacity to generate base or acid.
- It is therefore a fundamental soil property rather than a consequence of soil solution interaction.



# Pink Leach - Results

- High extractable mobile Al at all zones of non-neutrality.

Zones of acid generation correspond to zones of high Al extraction.

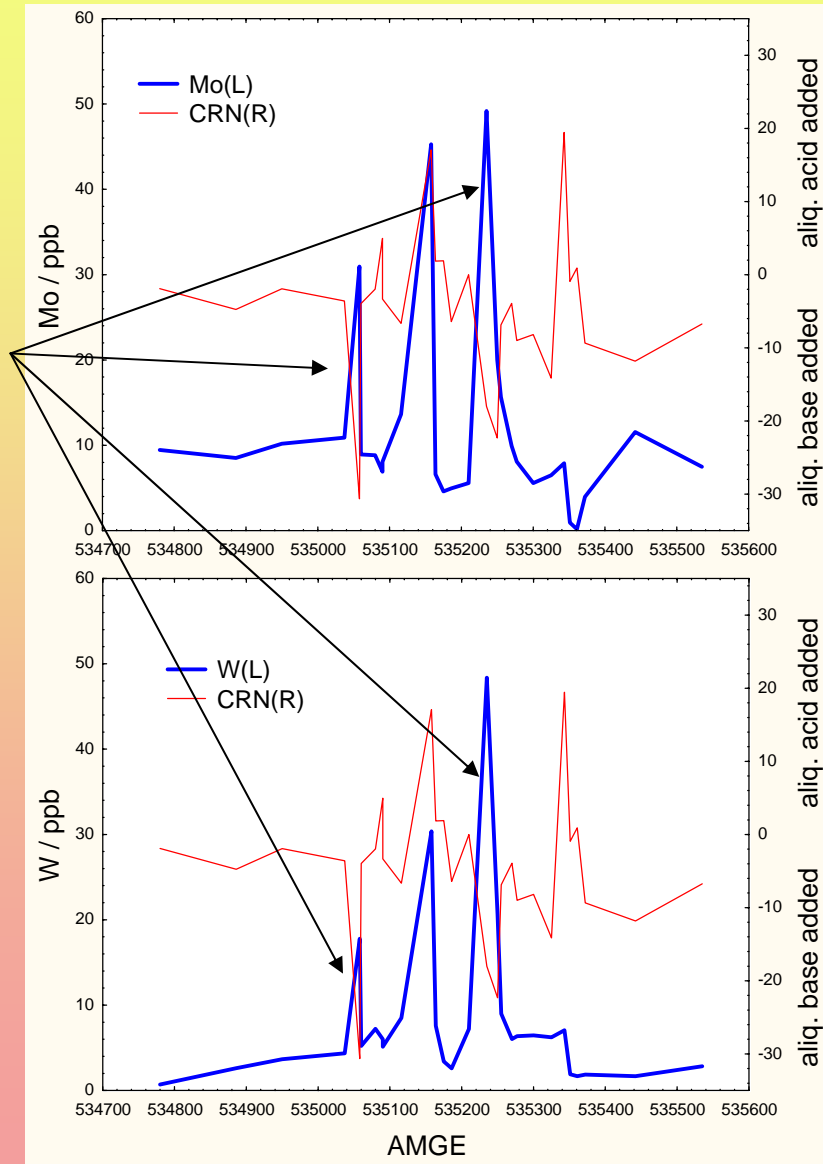


Zones of acid consumption correspond approximately to zones of high Al extraction.



# Pink Leach - Results

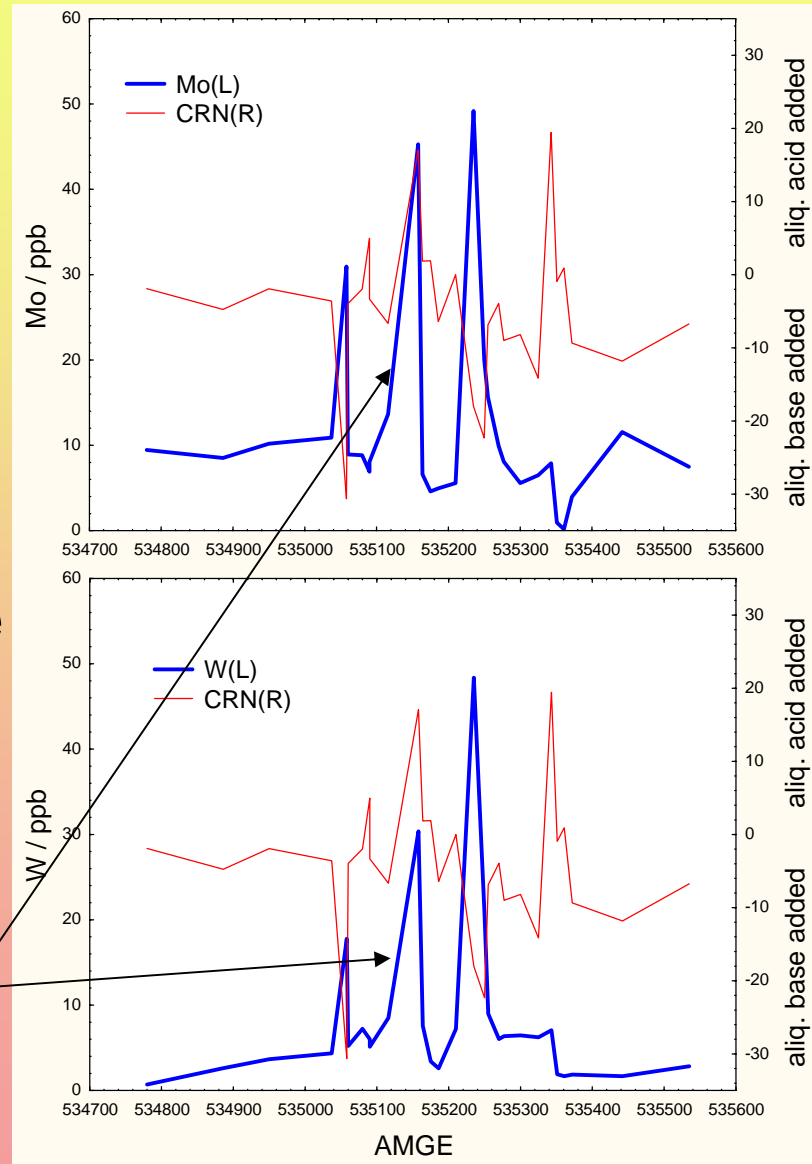
- Mo and W have a high extraction in the samples that generate acid.





# Pink Leach - Results

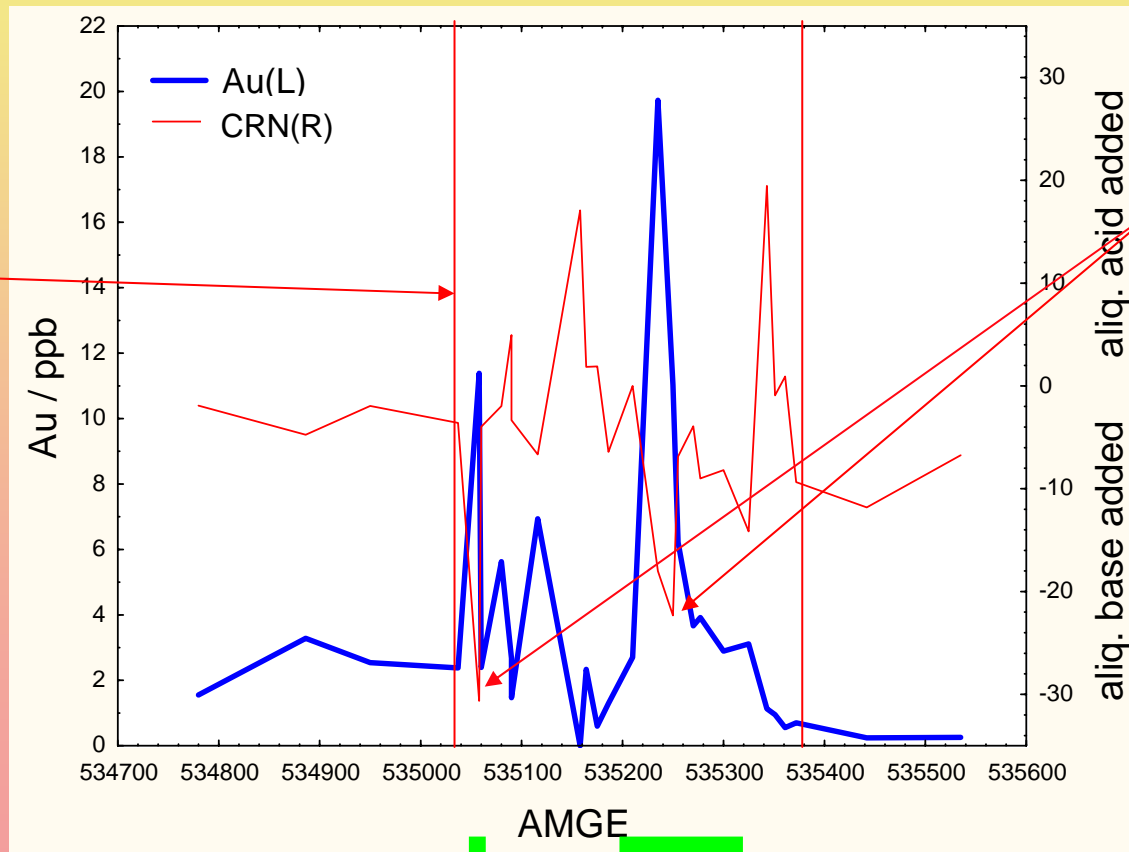
- Mo and W also accumulate in the acid consumption zone between the acid generation zones.
  - This may be a consequence of the limited directions for these elements to diffuse away.



# Pink Leach - Results

- Au has a high extraction in the samples which generate acid.

The high Au variability lies within the zone of non neutrality.

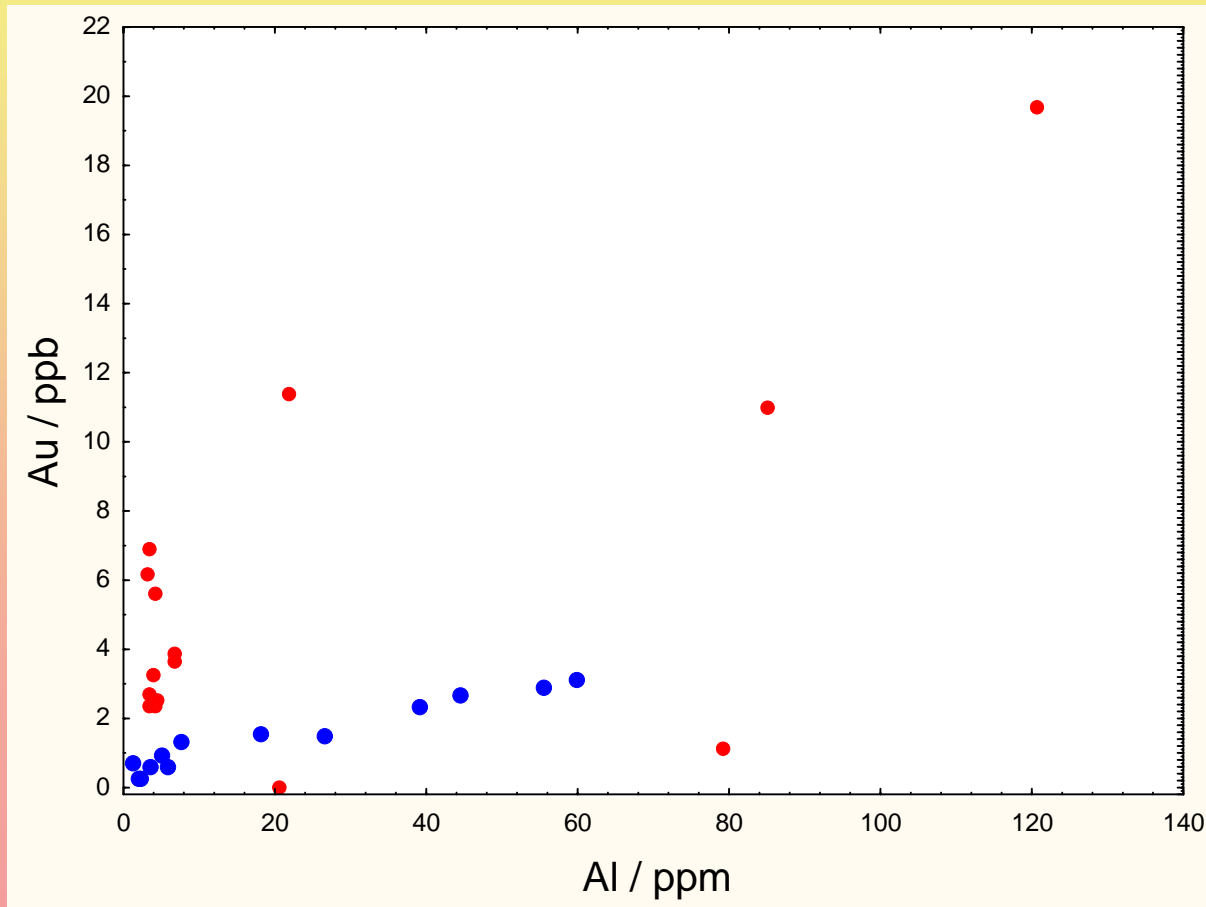


The major Au peaks correspond to the high acid generation samples.



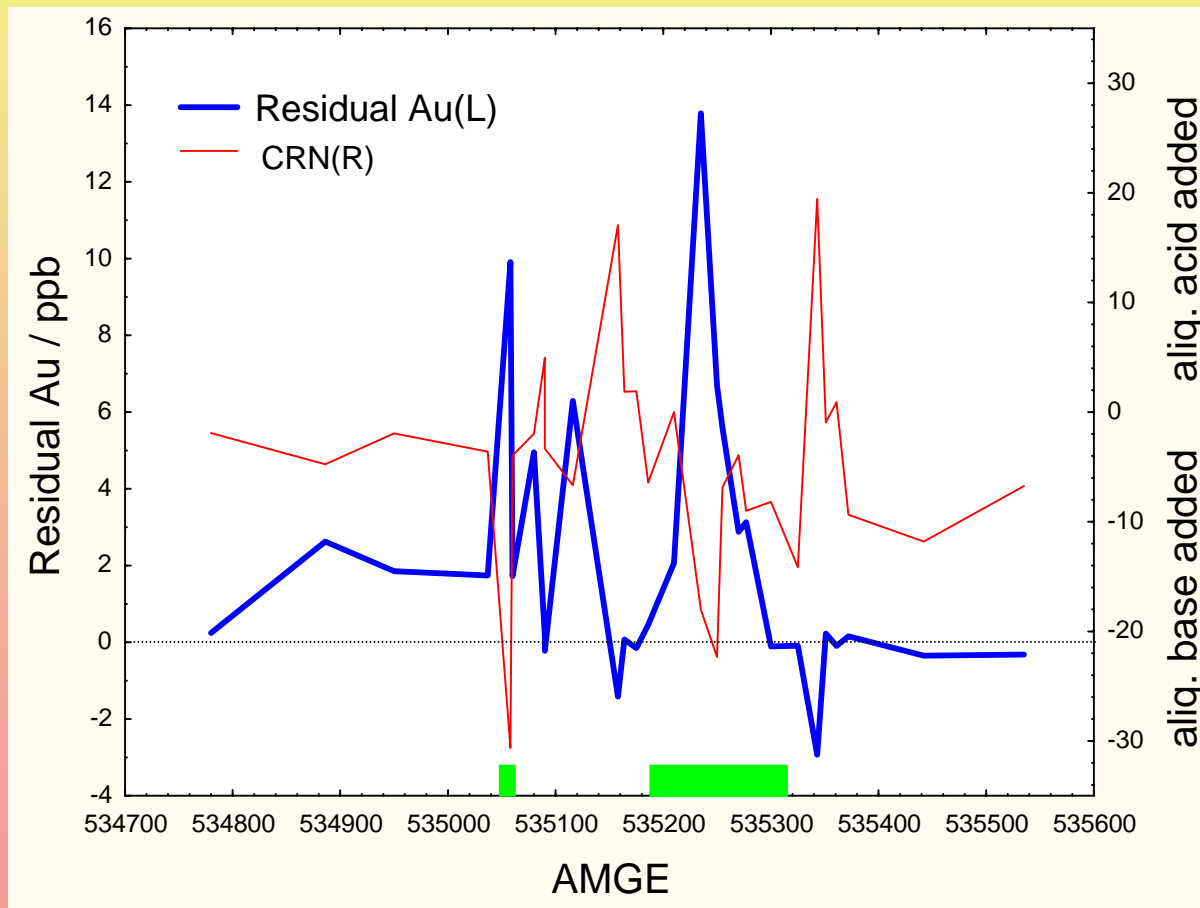
# Pink Leach - Results

- In a selection of samples (blue), the extraction of mobile Au is related mobile Al. Local residual Au minima correspond to the alkaline generation zones.



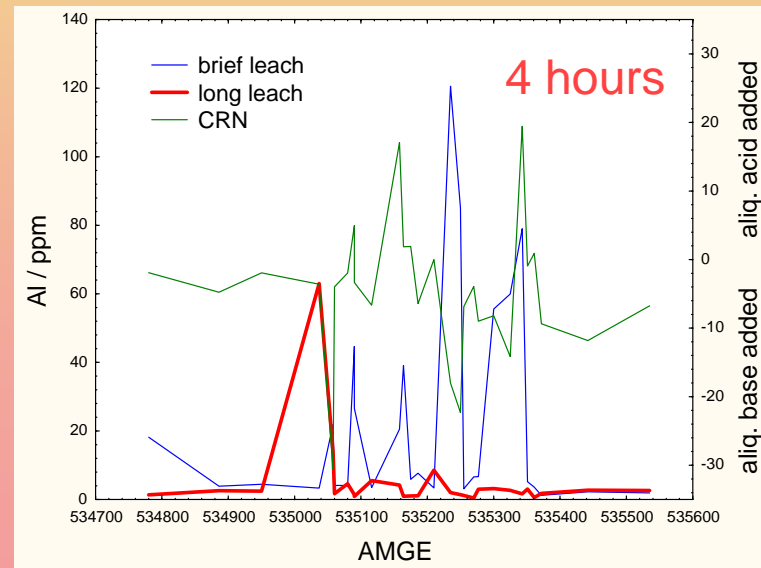
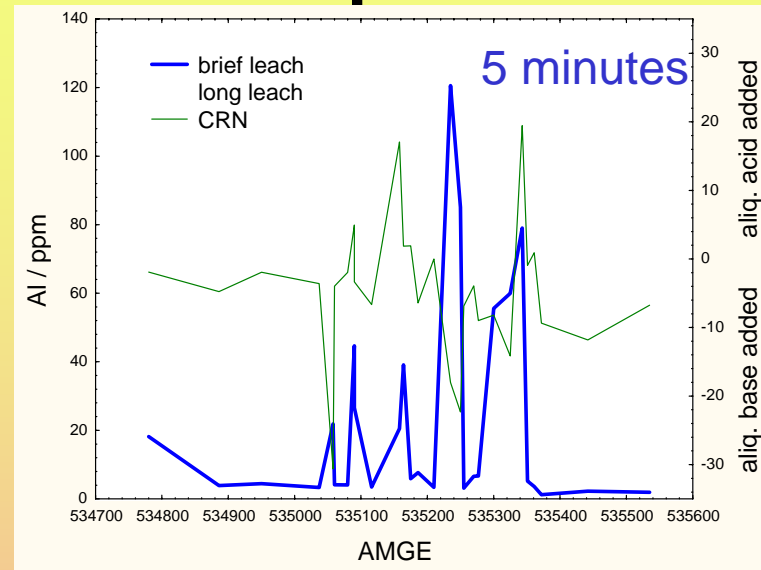
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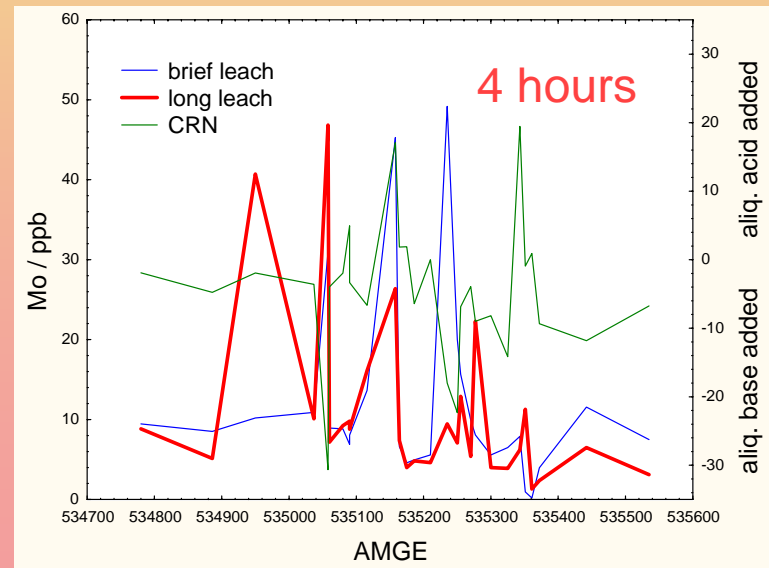
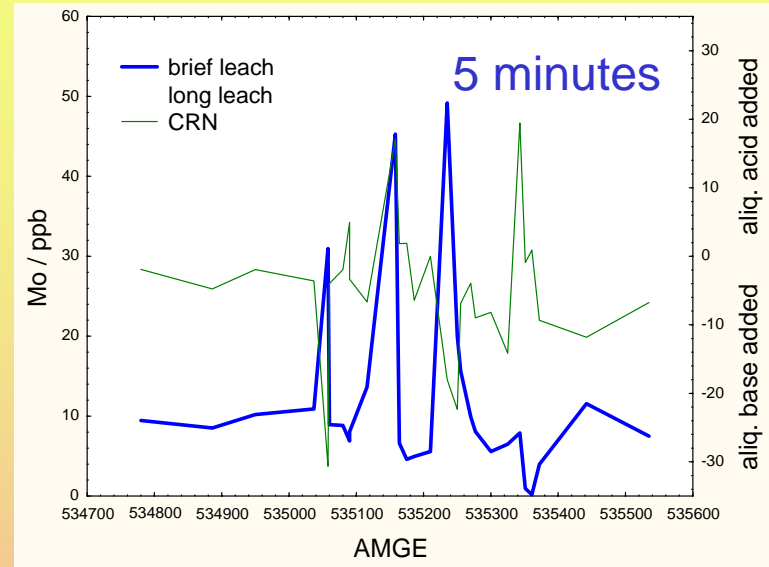
# Pink Leach – Time Dependence

- Al quickly destabilises out of solution.
- Au, Mo and W also destabilise quickly out of solution, particularly over mineralisation. Patterns are obscured in a high background.



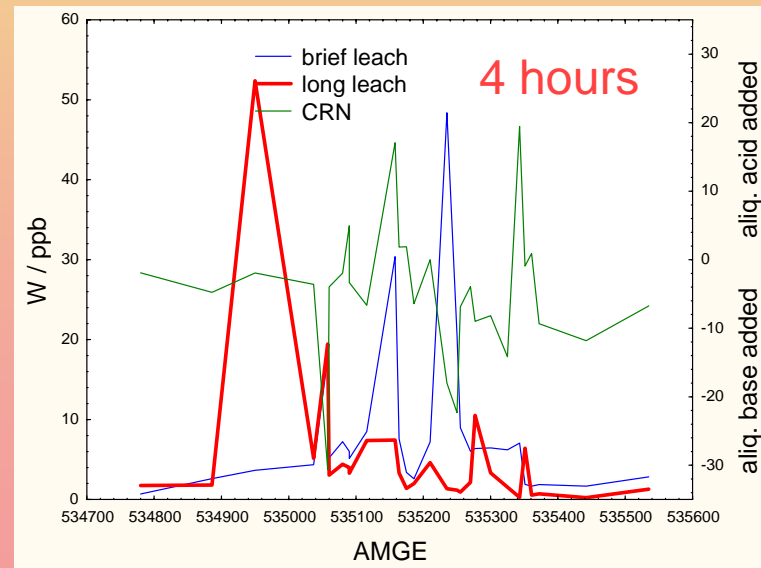
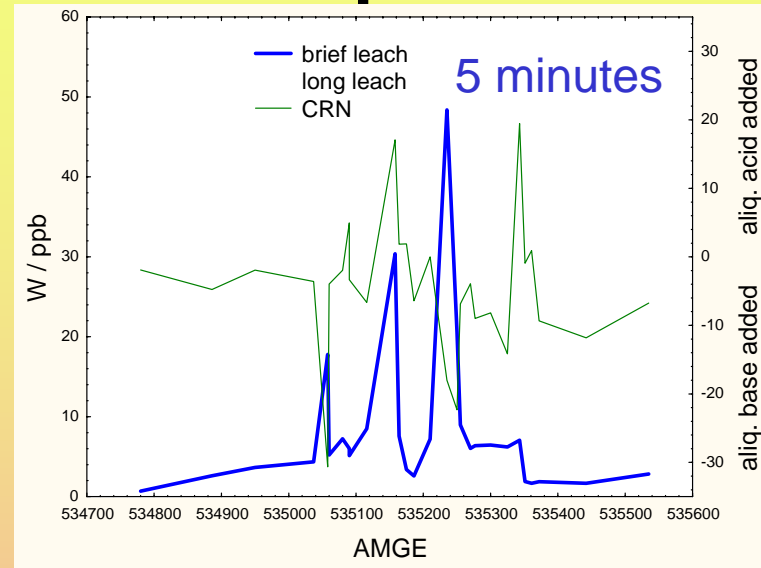
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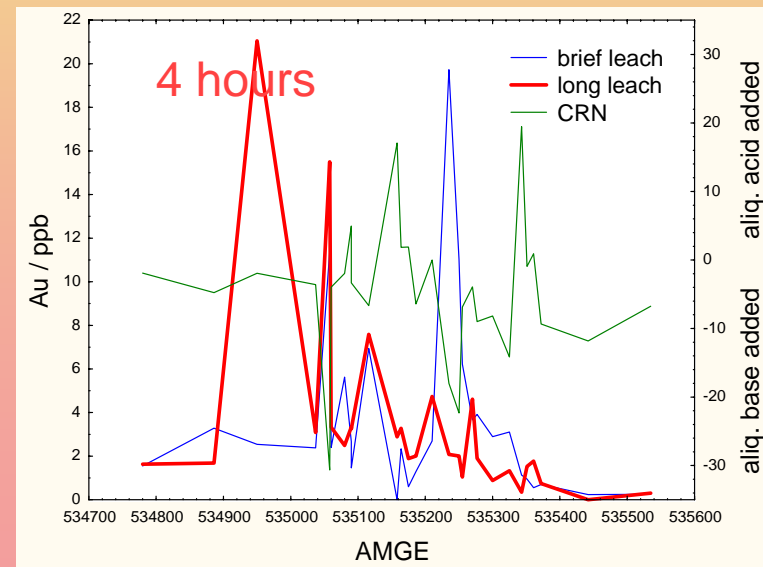
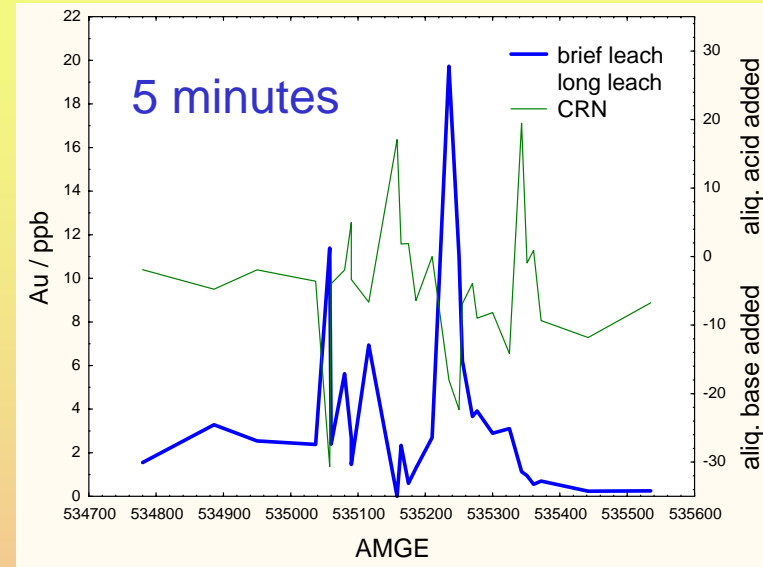
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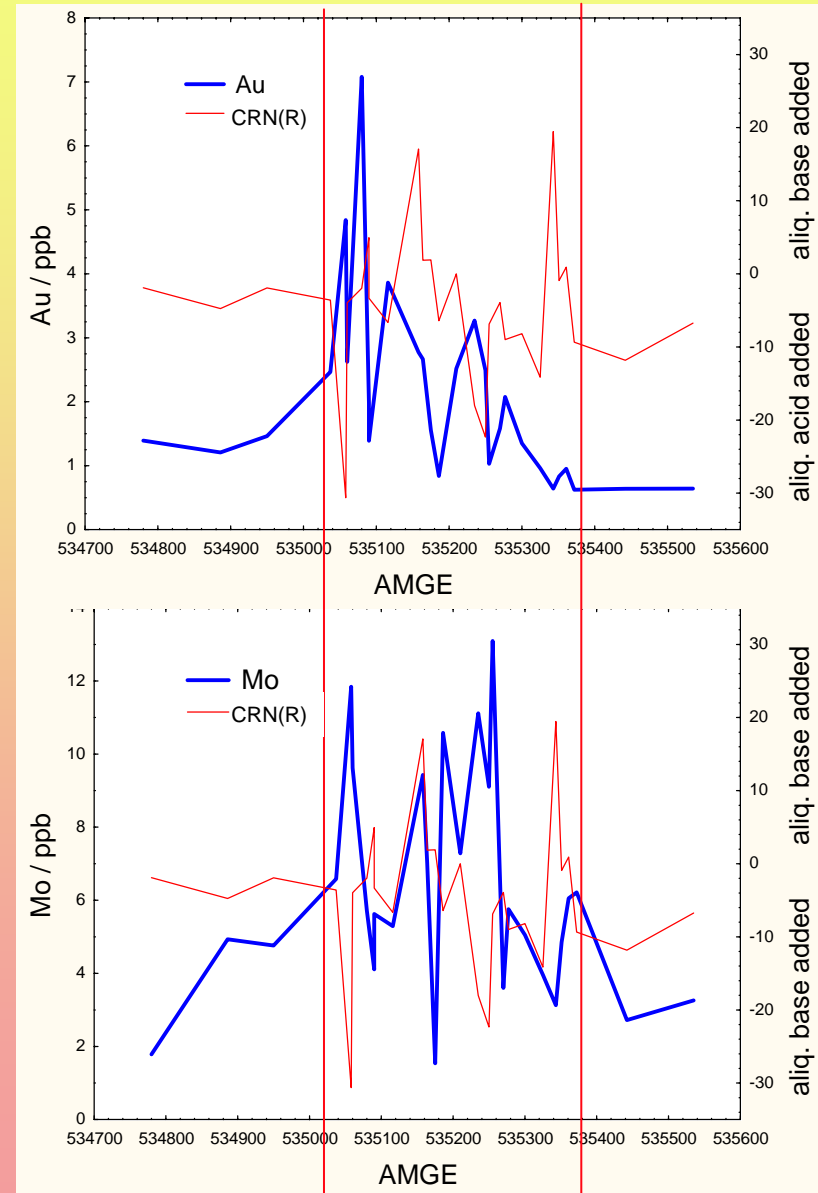
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# Pink Leach – Other Leaches

- The local acid-base conditions can be used to interpret the results other leaches.
- Fine detail which might otherwise be interpreted as variability is related to the local acid-base system.



# Pink Leach

- Local acid base variations are efficiently illustrated by the capacity of a sample to consume or generate acid.
- The theory behind this leach assumes that local acid base variations may be related to the movement of acid from an oxidising ore body at depth through local structures.
- This subtle leach appears to simply reverse these acid base processes.



# Pink Leach

- If the background pH is different from 7.5 or a different suite of elements was targeted, an alternative indicator could be used.
- The acid base consumption can be used as an inexpensive field technique to screen for potential partial leach responses.



# Pink Leach - Conclusions

- An understanding of how an individual sample location relates to the local geochemical system enhances the interpretability of the partial leach results.
- An anomalous response related to mineralisation was seen at Mandamah, where other techniques had failed.

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