

24th IAG Symposium
Fredericton, Canada

Workshop B: Indicator Mineral Exploration Technology

Viable indicators in surficial sediments for two major base metal deposit types: Ni-Cu-PGE and porphyry Cu

Presented by Stu Averill

OVERBURDEN DRILLING MANAGEMENT LIMITED

May 31, 2009



Selected Properties of Indicator Minerals

1. Heavy (due to low abundance)
2. Coarse-grained (>0.25 mm; unless ultra-heavy – e.g. gold and PGMs)
3. Resistant to weathering (eliminates most sulphide minerals; increases dependence on alteration minerals)

High pressures and temperatures increase mineral density and grain size

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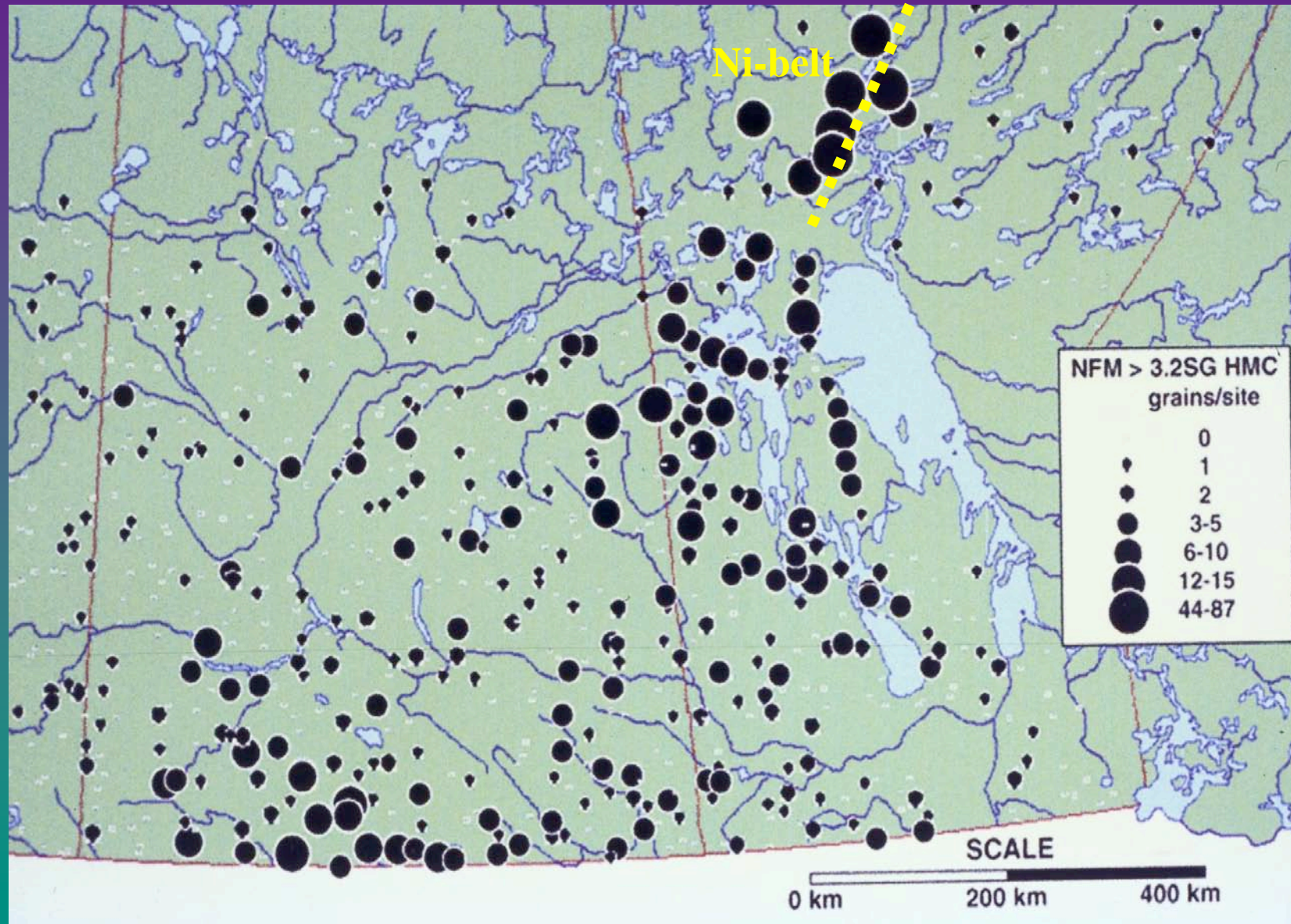
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Dispersal of Cr-diopside from Thompson Ni-Belt



Courtesy: Harvey Thorlietson



There are different indicator mineral suites and subsuites for:

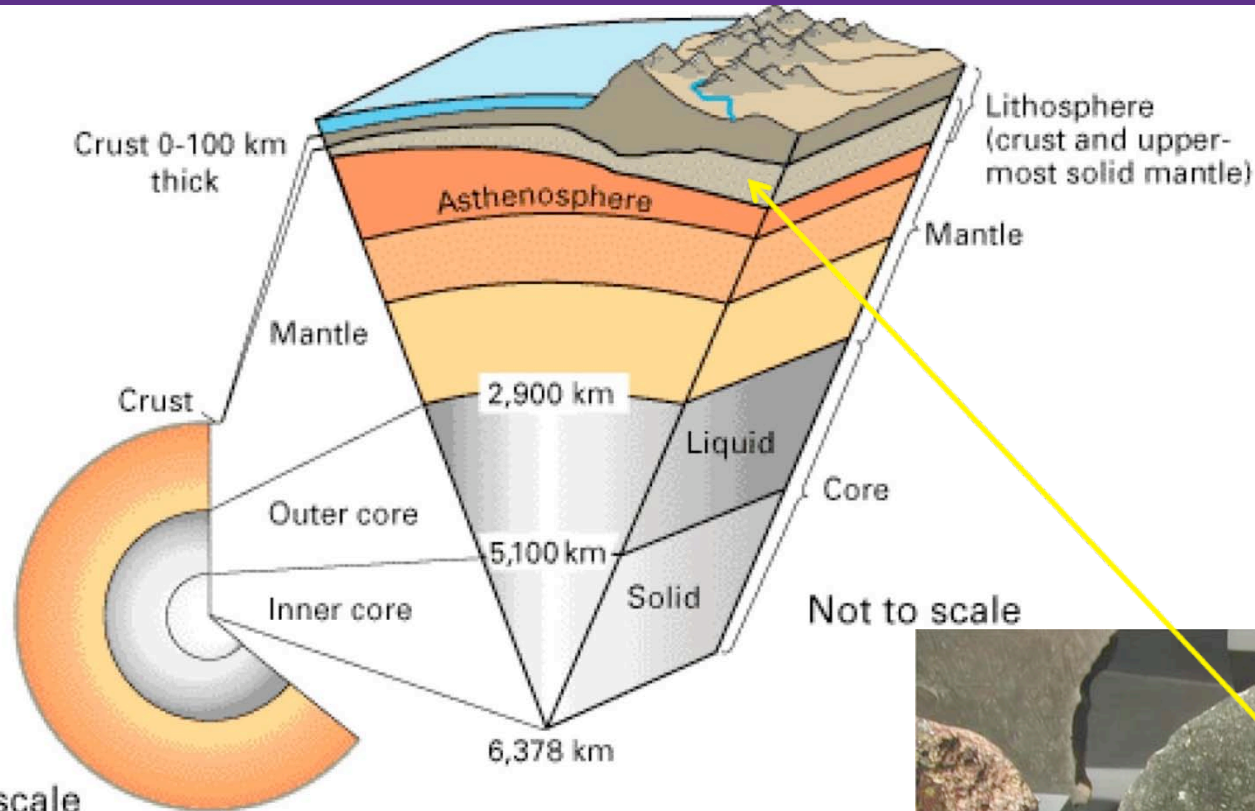
- Ni-Cu-PGE and porphyry Cu deposits
- the successive mineralizing events or processes that form these deposits
- each hydrothermal alteration zone

Outline 1 – Ni-Cu-PGE Indicator Minerals

Four mineral subsuites indicating:

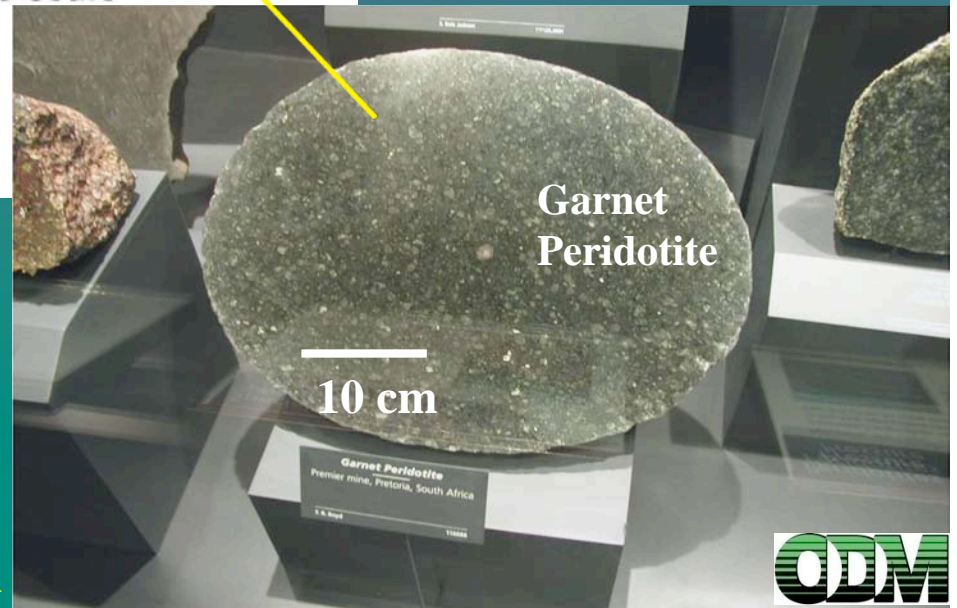
1. a fertile melt
2. rapid, localized fractionation of cumulus minerals from the melt (promotes sulphide saturation)
3. assimilation of felsic rocks by the melt (also promotes sulphide saturation)
4. actual mineralization

The Garnet Peridotite Connection for Ni-Cu-PGE and Kimberlite Indicators



To scale

Courtesy: Bruce Kjarsgaard, GSC



Courtesy: Smithsonian Institution



Outline 1 – Ni-Cu-PGE Indicator Minerals

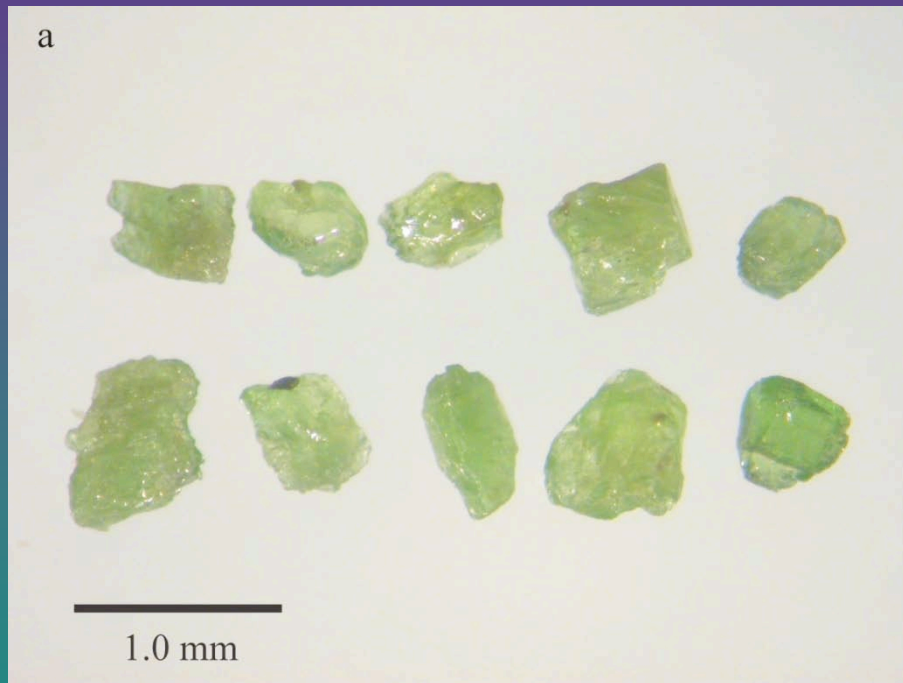
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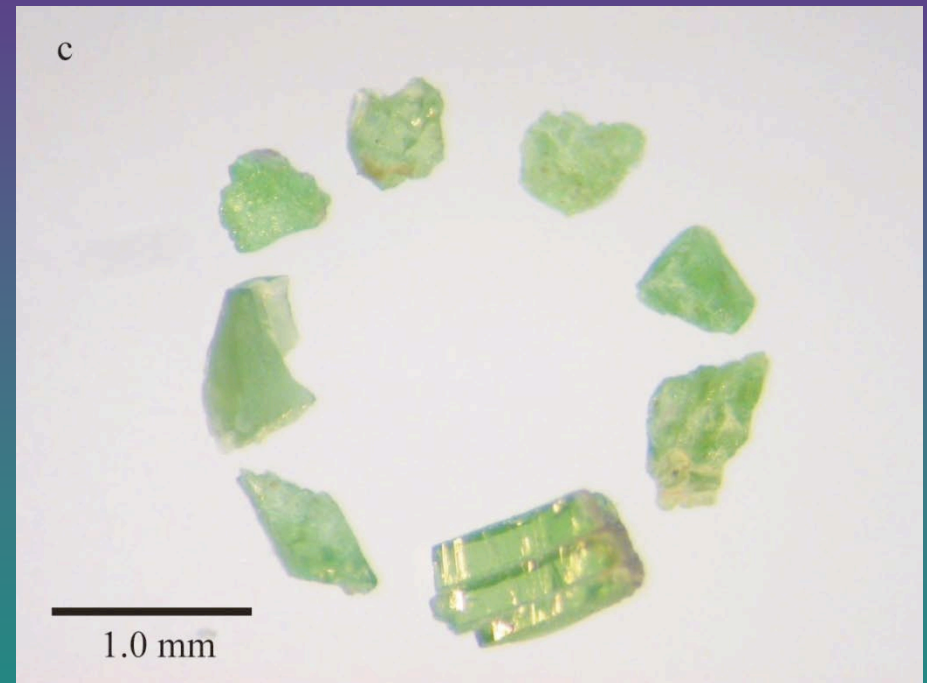
Indicators of a Fertile Melt

- orthopyroxene (enstatite – $\text{Mg}_2\text{Si}_2\text{O}_6$)
- olivine (forsterite – MgSiO_4)
- Cr-diopside – $\text{Ca}(\text{Mg},\text{Cr})\text{Si}_2\text{O}_6$
- chromite – $(\text{Fe},\text{Mg})(\text{Cr},\text{Al})\text{O}_4$

Cr-diopside

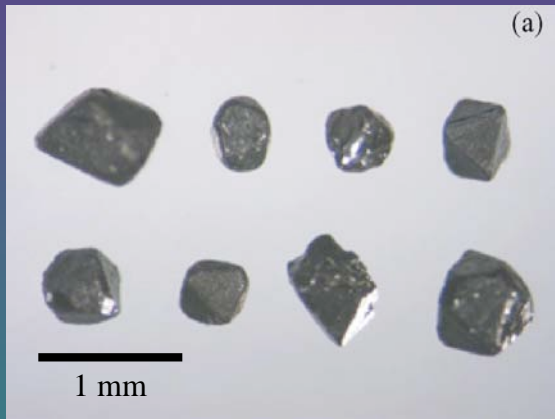


Non-kimberlitic
<1.25% Cr₂O₃

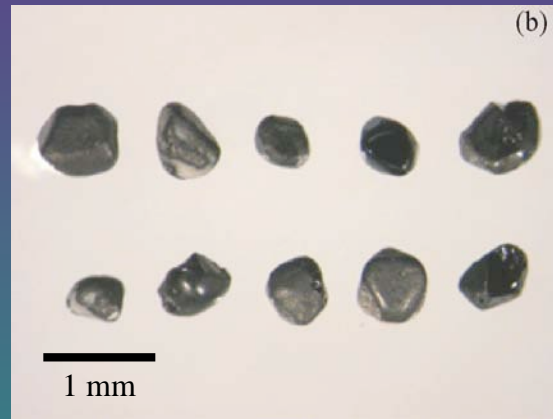


Kimberlitic
>1.25% Cr₂O₃

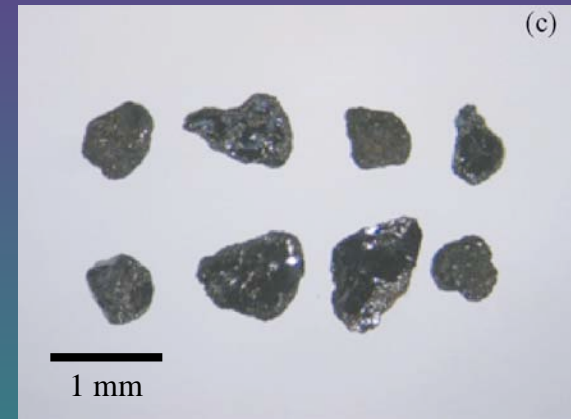
Chromite



Non-kimberlitic



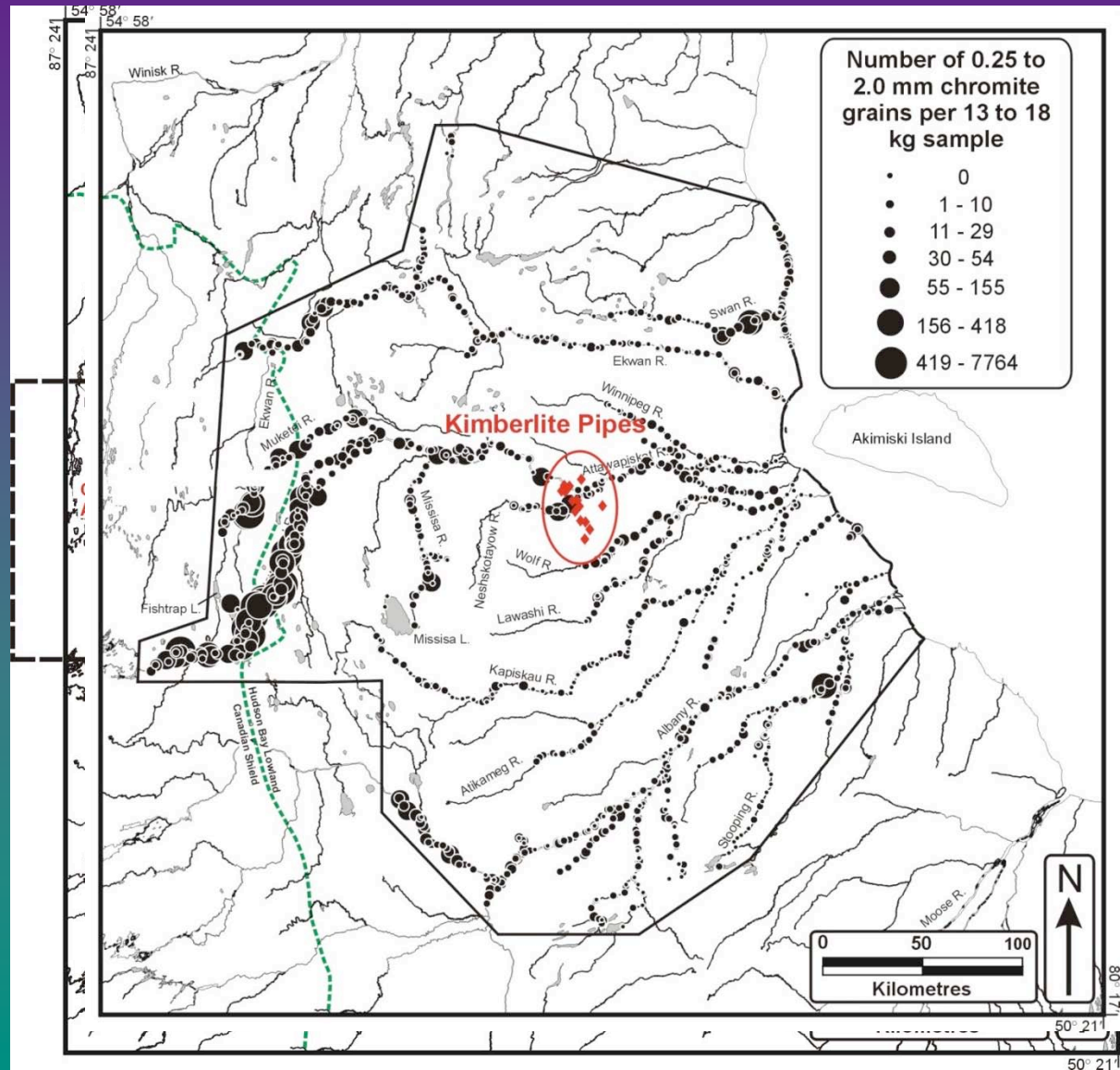
Kimberlitic



Lateritic

c

Chromite Dispersal in the Attawapiskat River



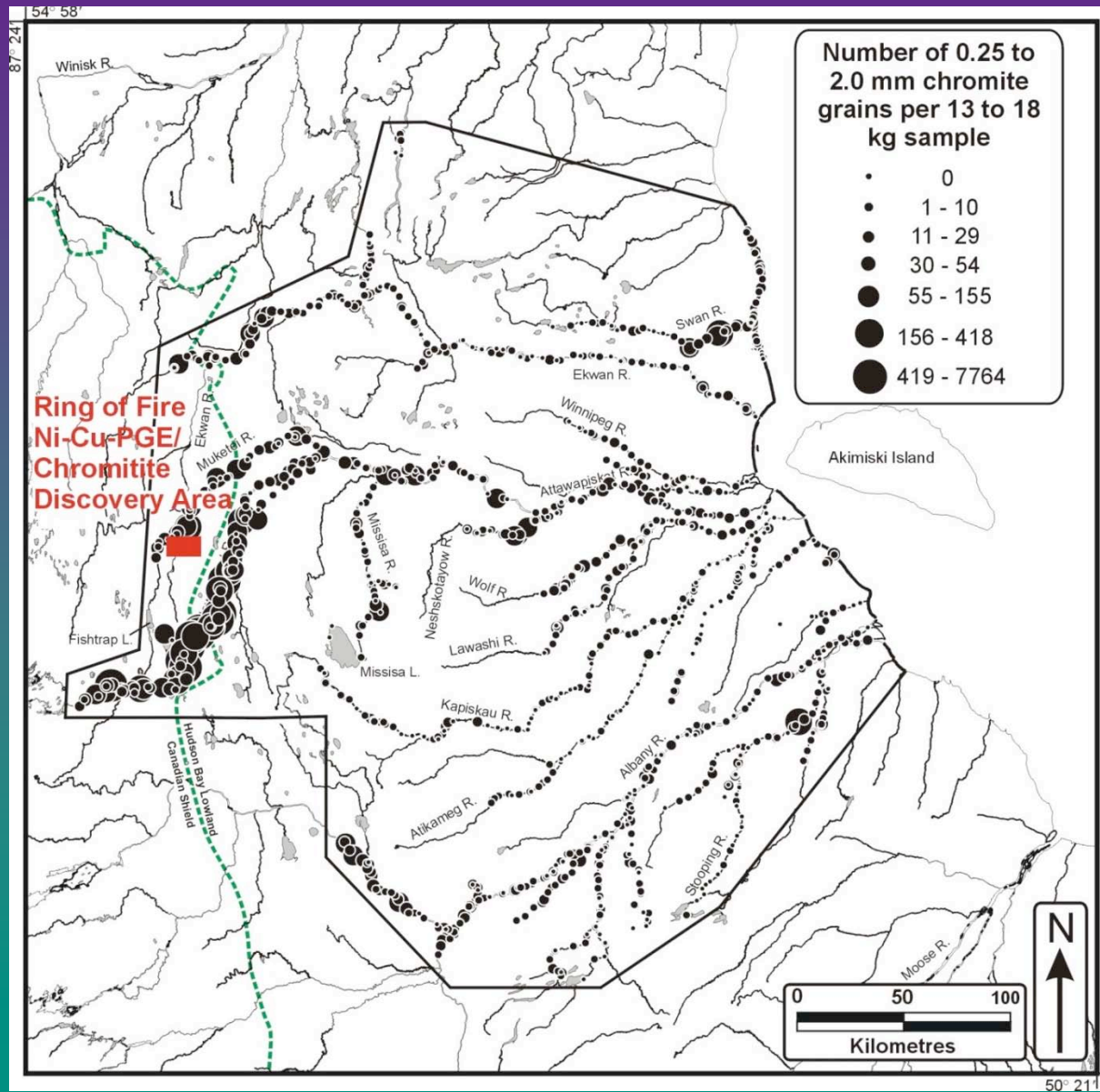
Modified from: Ontario Geological Survey

Chromitite Grains, Attawapiskat River



1.0 mm

“Ring of Fire” Chromitite Discoveries



Modified from: Ontario Geological Survey



70 m of Massive Chromitite



Courtesy: Noront Resources Ltd.

Role of Sulphide Saturation

- Causes sulphide liquid to separate from silicate melt
- Sulphide liquid collects Ni-Cu-PGE from silicate melt
- Heavy sulphide liquid settles in pools or layers, further concentrating metals

Outline 1 – Ni-Cu-PGE Indicator Minerals

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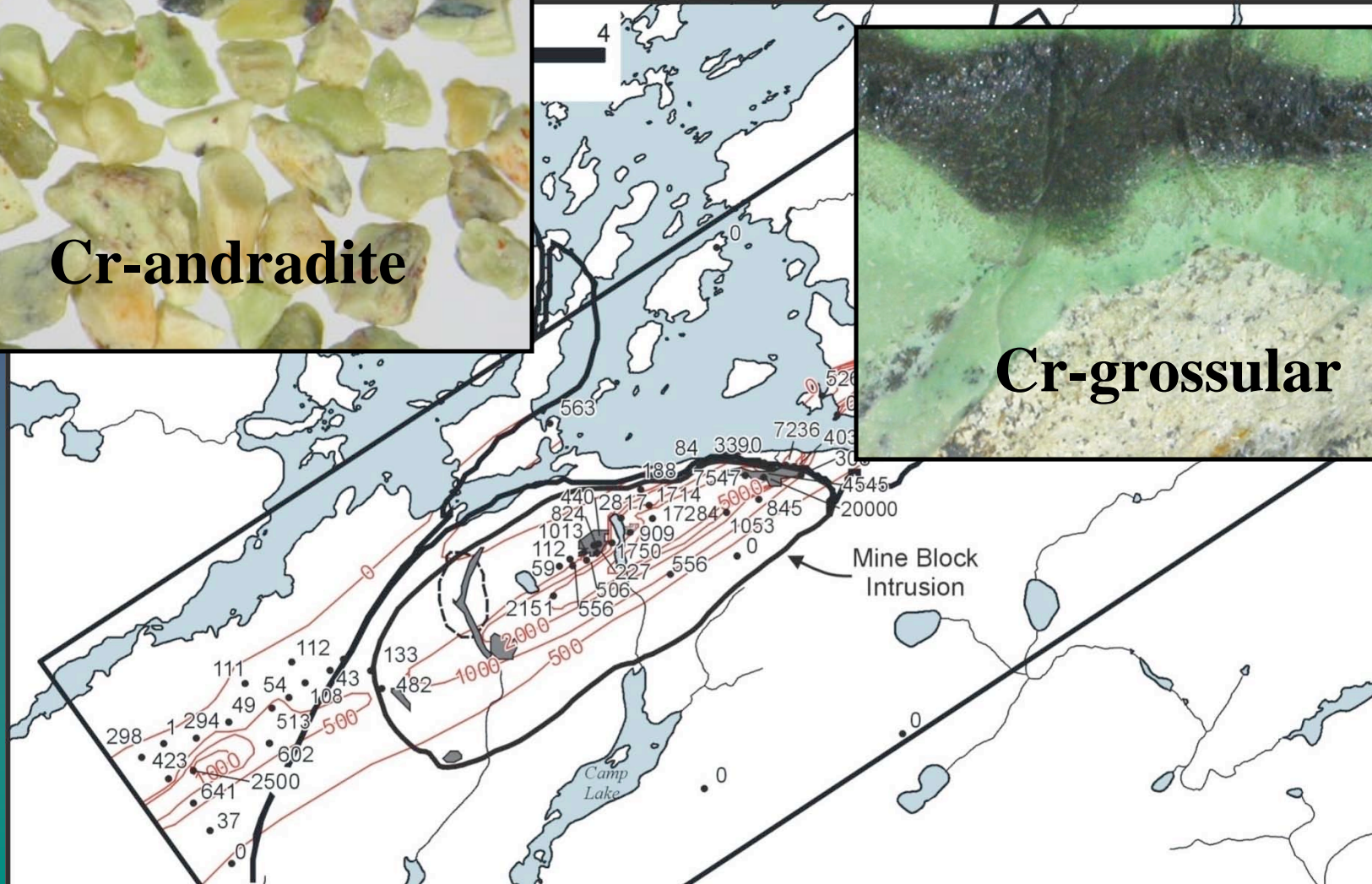
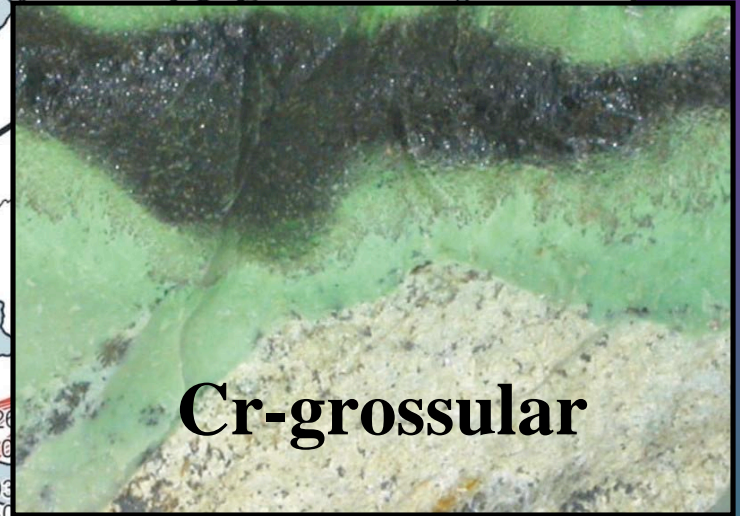
Indicators of Concentrated Cumulus Segregation

- orthopyroxene (enstatite – $\text{Mg}_2\text{Si}_2\text{O}_6$)
- olivine (forsterite – MgSiO_4)
- Cr-diopside – $\text{Ca}(\text{Mg},\text{Cr})\text{Si}_2\text{O}_6$
- chromite – $(\text{Fe},\text{Mg})(\text{Cr},\text{Al})\text{O}_4$

Ruby Corundum $(\text{Al,Cr})_2\text{O}_3$



Dispersal of Cr-andradite from Lac des Iles Intrusive Complex



Outline 1 – Ni-Cu-PGE Indicator Minerals

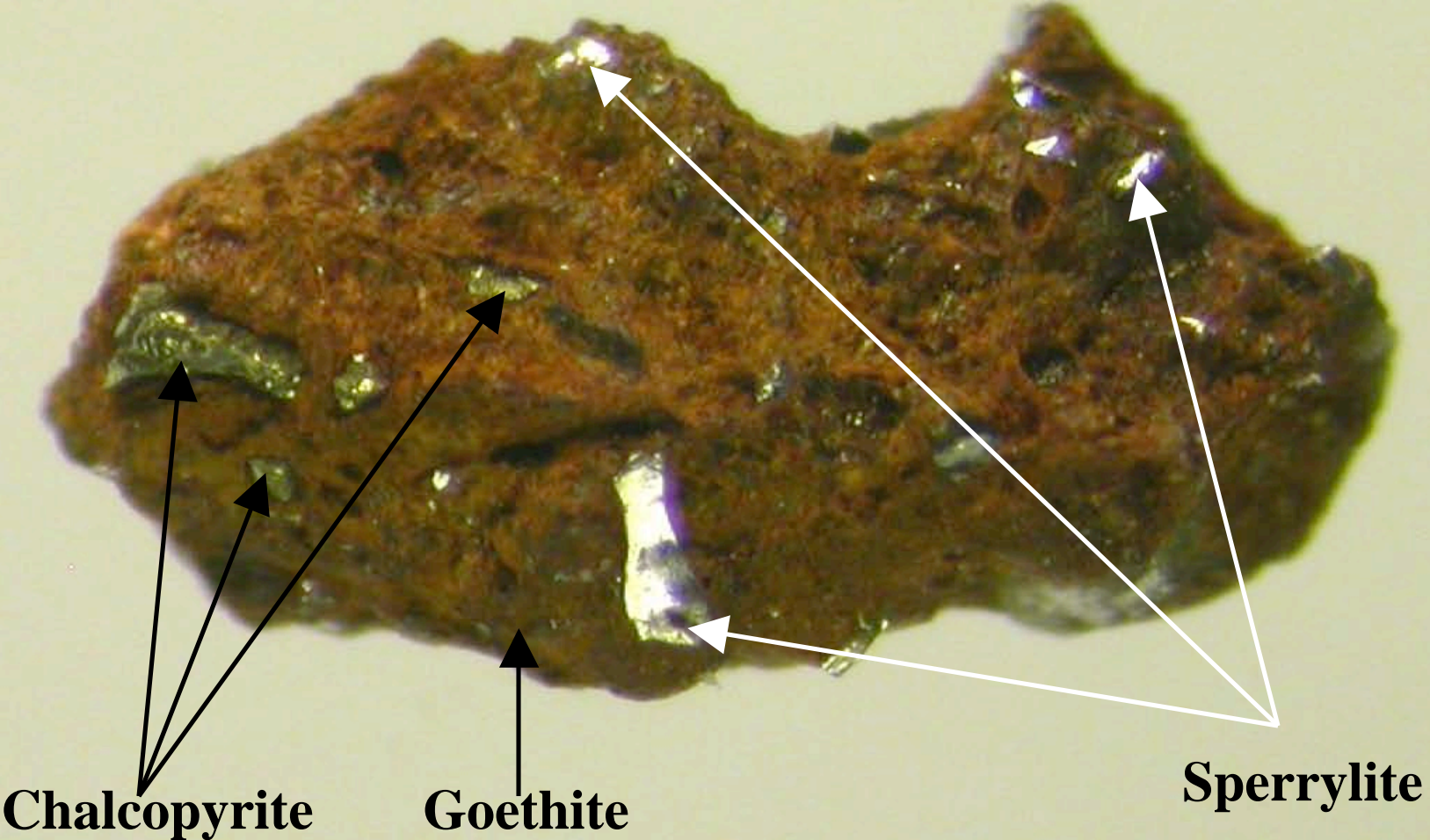
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Relative stabilities of Fe-sulphides and Ni-Cu-PGE ore minerals

<u>Mineral</u>	<u>Stability</u>
Ni-sulphides	unstable
PGE-sulphides	unstable
PGE-tellurides	unstable
pyrrhotite	unstable
pyrite	unstable
chalcopyrite	significantly stable
FeNi and PGE-arsenides	stable (but silt-sized)
PGE-antimonides	stable (but silt-sized)
native Au and PGE	very stable (but silt-sized)

Broken Hammer Gossan



Chalcopyrite

Goethite

Sperrylite

1 mm

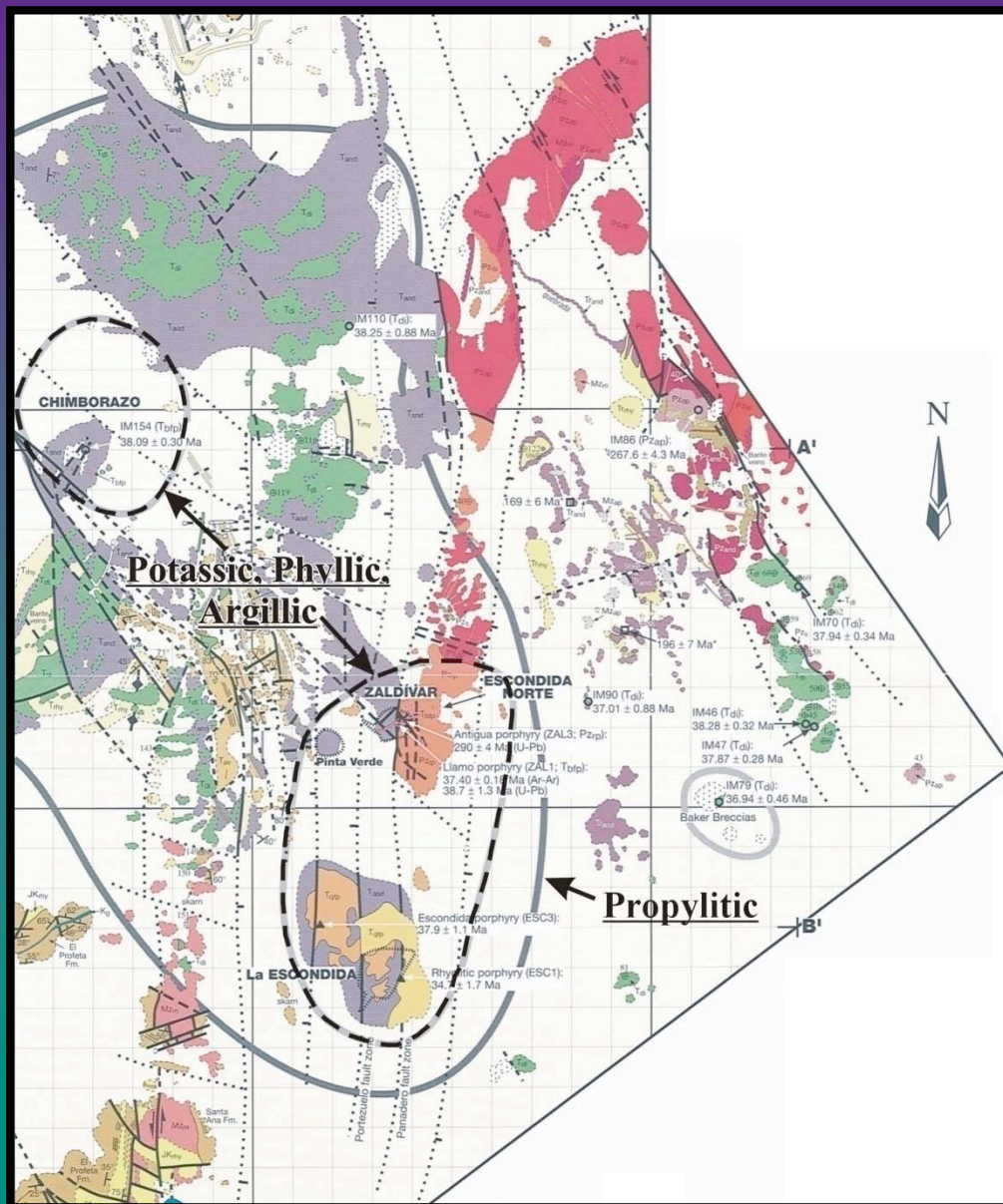


Outline 2 – Porphyry Cu Indicator Minerals (PCIMs[®])

PCIM[®] anomalies, like Ni-Cu-PGE anomalies are:

1. Big
2. Strong
3. Zoned

Alteration zones, Escondida, Chile



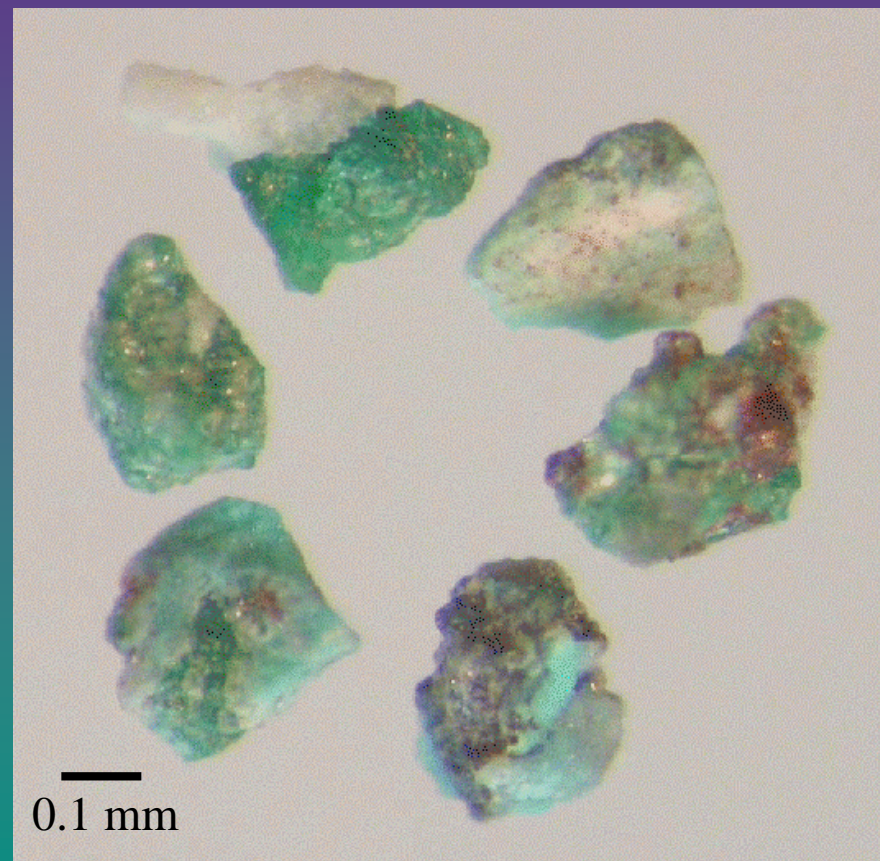
Arid landscape, Atacama Desert, Chile



Jarosite



Atacamite



Mineral	Density	Composition	Principal provenance (alteration zone)				
			Potassic	Argillic	Phyllic	Propylitic	Epithermal Au

Hypogene suite:

Diaspore	3.4	AlO(OH)			=====		
Alunite	2.9	(K,Na)Al ₃ (SO ₄) ₂ (OH) ₆	=====				
Dravite	3.0	NaMg ₃ Al ₆ (BO ₃) ₃ (Si ₆ O ₁₈)(OH) ₄			=====		
Andradite	3.9	Ca ₃ Fe ₂ (SiO ₄) ₃				=====	
Barite	4.5	BaSO ₄					=====

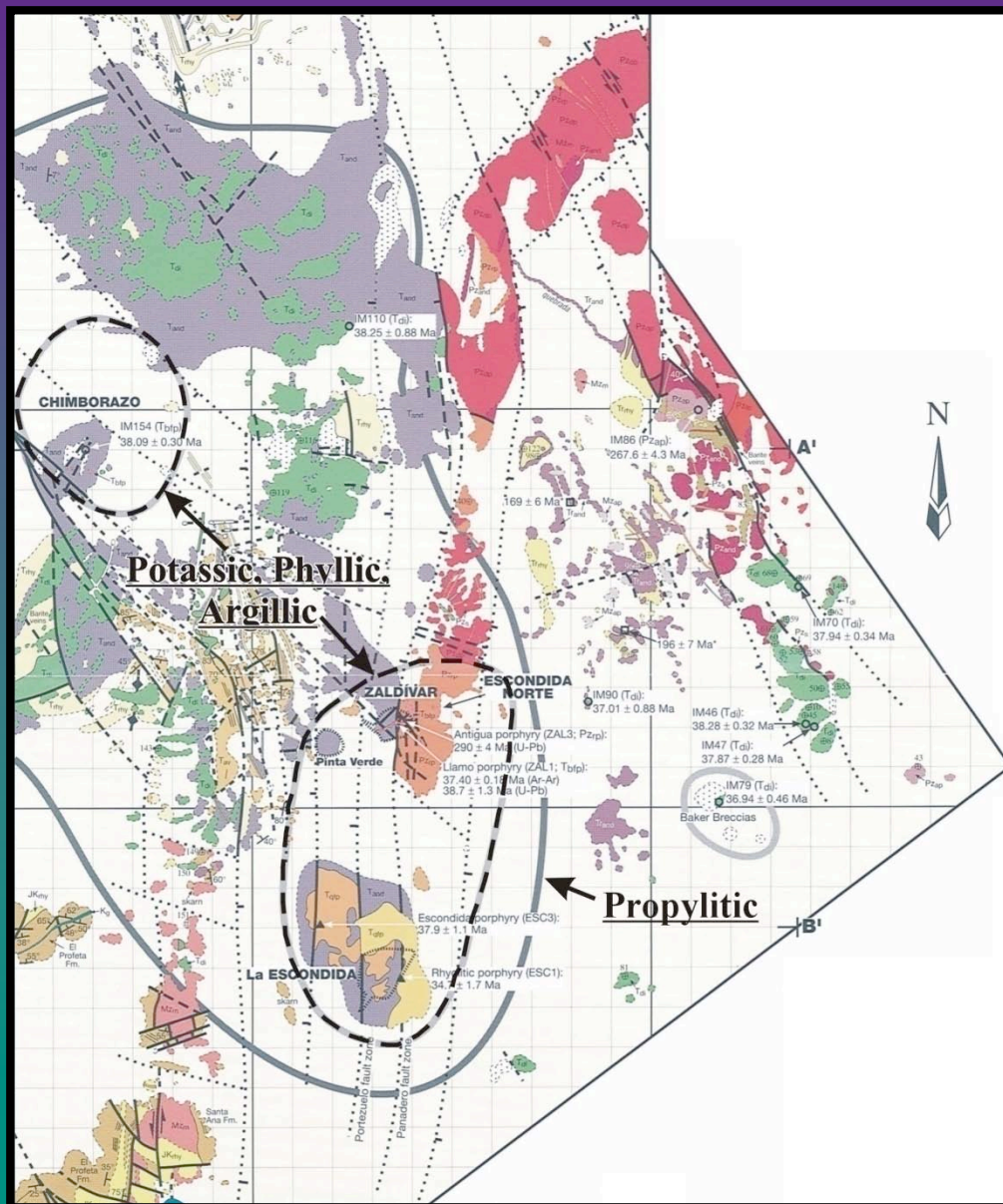
Supergene suite:

Alunite	2.8	(K,Na)Al ₃ (SO ₄) ₂ (OH) ₆	=====				
Jarosite	3.1	(K,Na)Fe ₃ (SO ₄) ₂ (OH) ₆	=====				
Atacamite	3.8	Cu ₂ Cl(OH) ₃			=====		
Turquoise	2.8	CuAl ₆ (PO ₄) ₄ (OH) _{8.5} H ₂ O				=====	
Malachite	4.0	Cu ₂ CO ₃ (OH) ₂					=====

Proven porphyry Cu indicator minerals (PCIMs[®])



Alteration zones, Escondida, Chile



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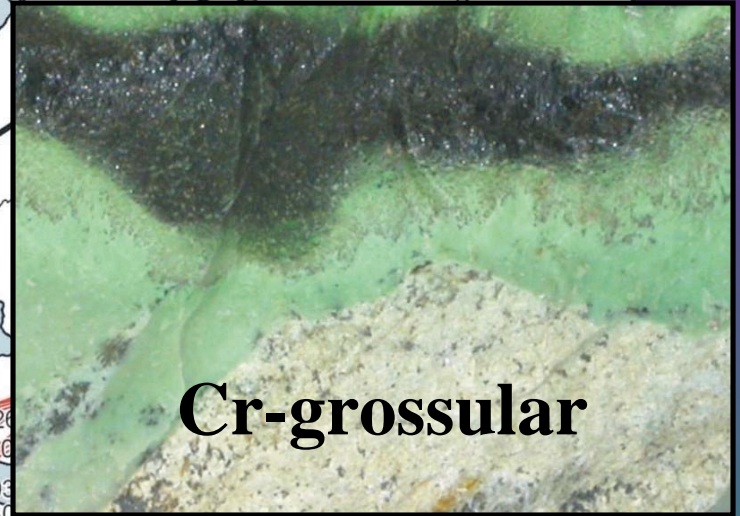
Proven porphyry Cu indicator minerals (PCIMs[®])



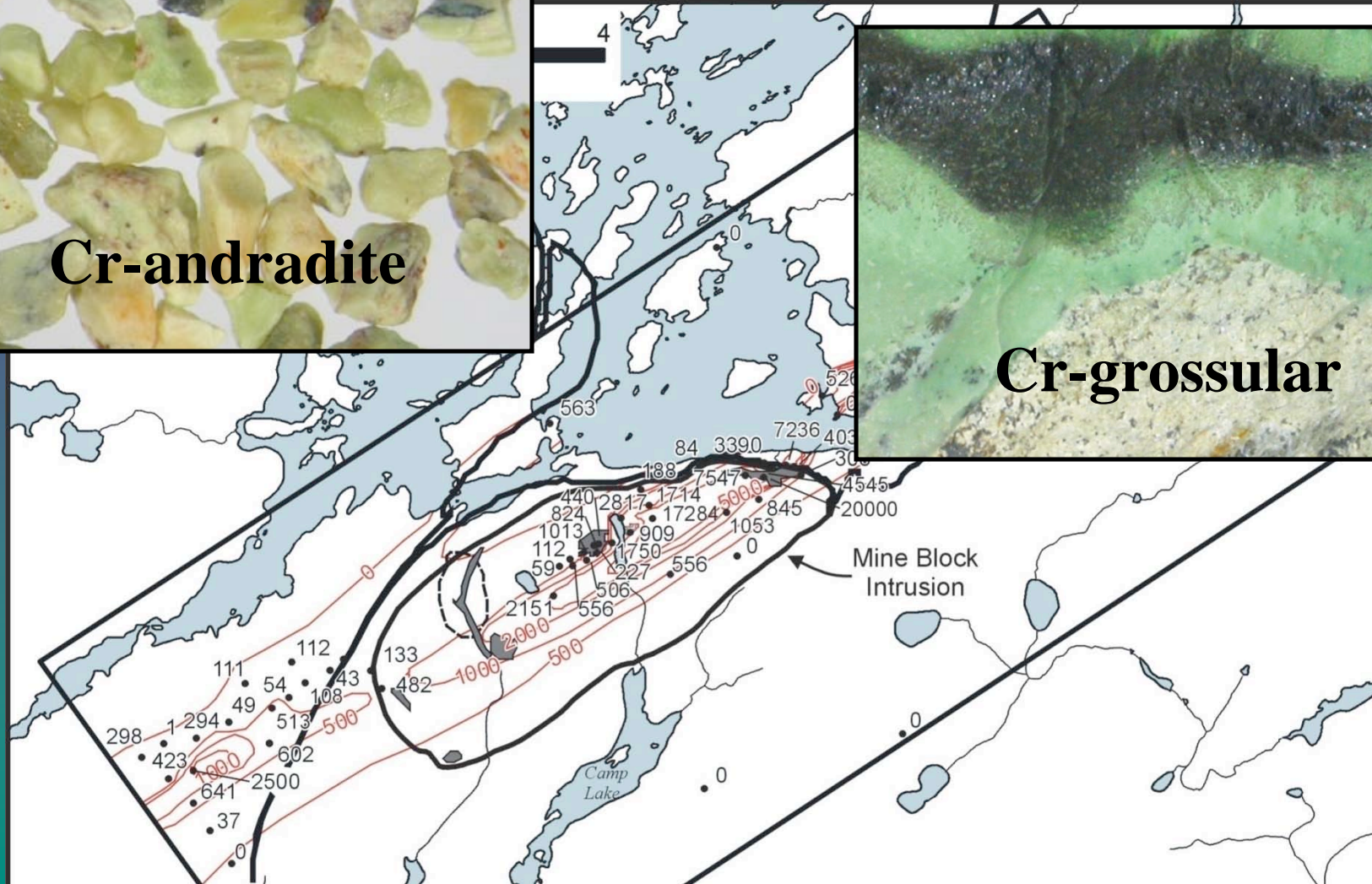
Dispersal of Cr-andradite from Lac des Iles Intrusive Complex



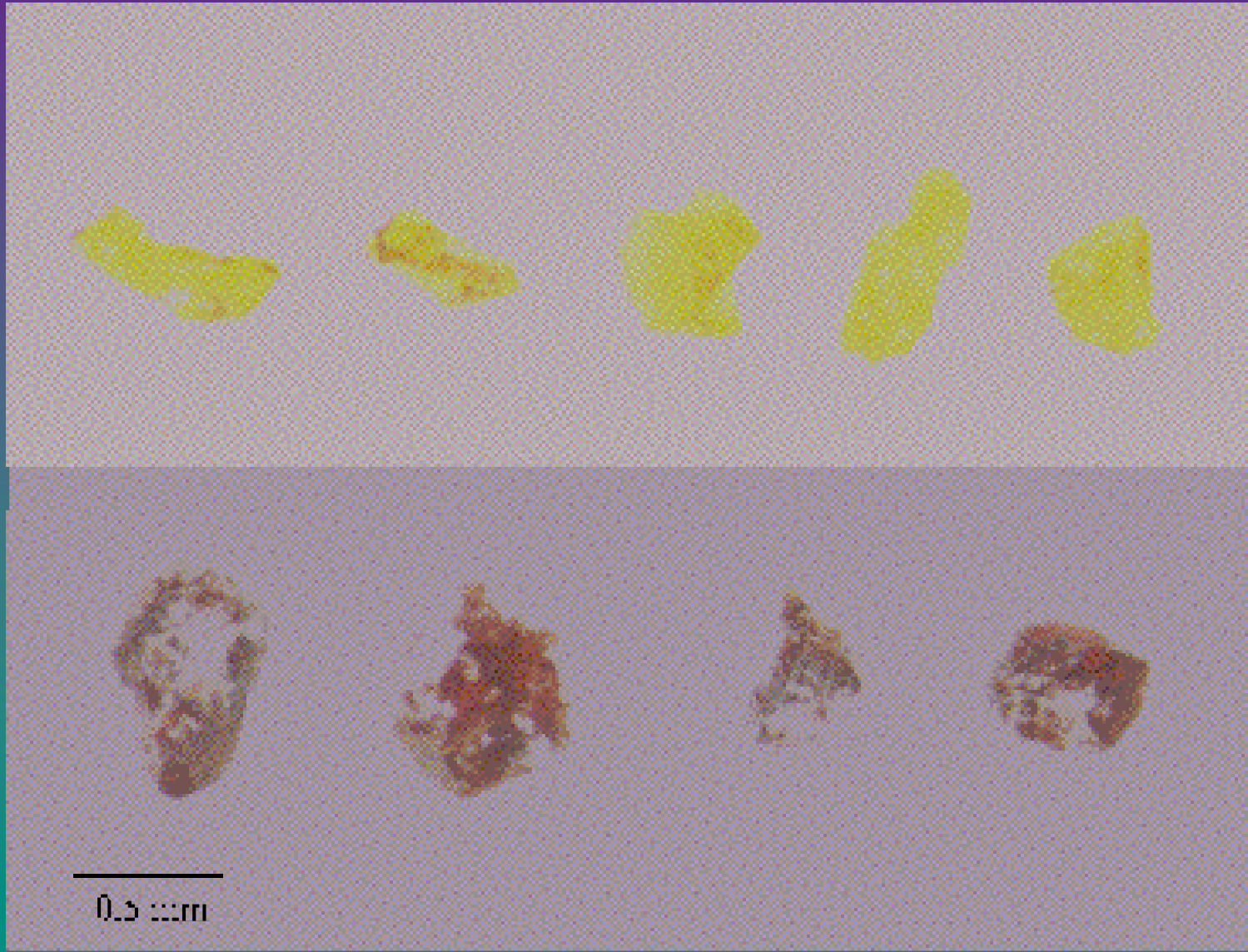
Cr-andradite



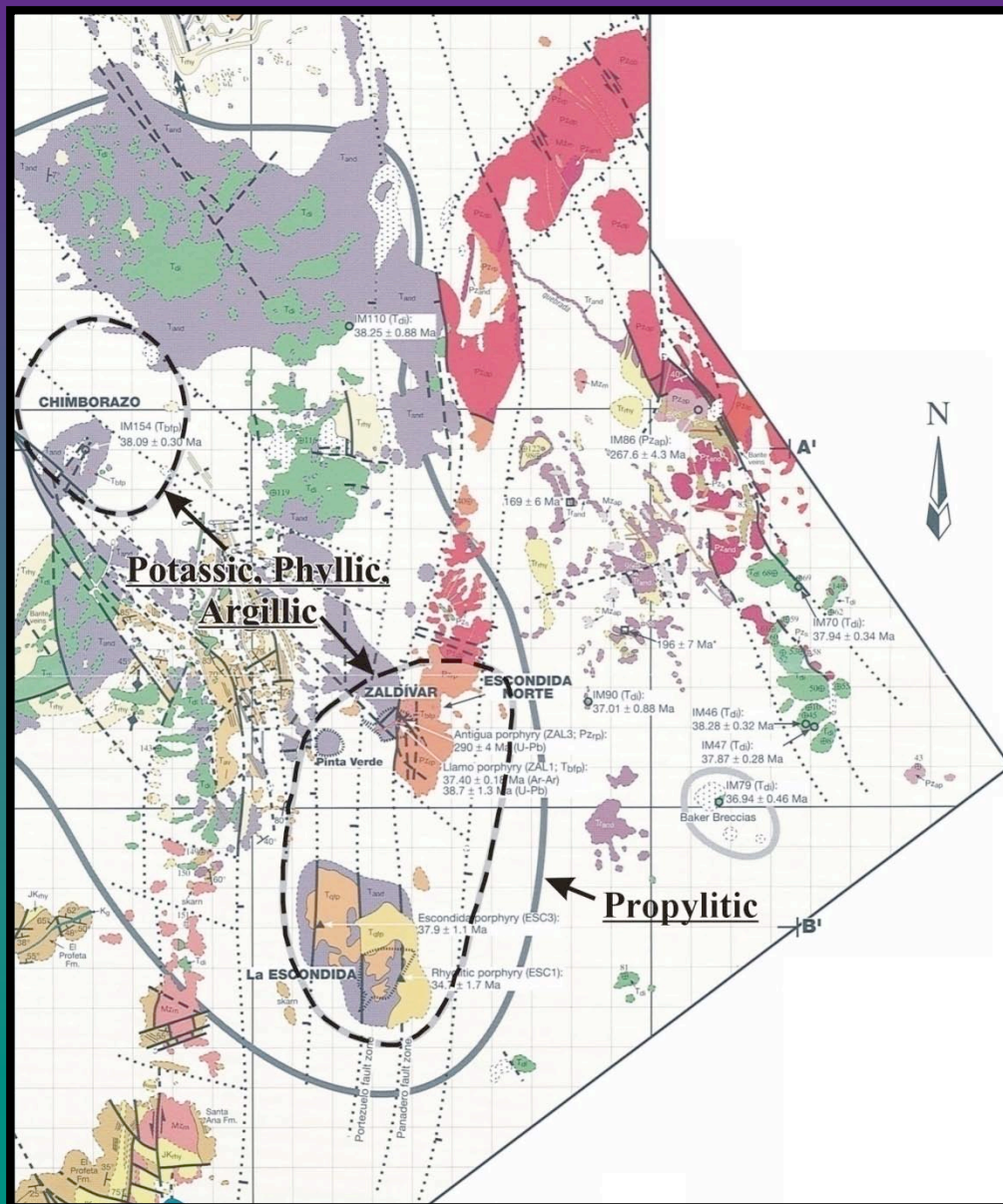
Cr-grossular



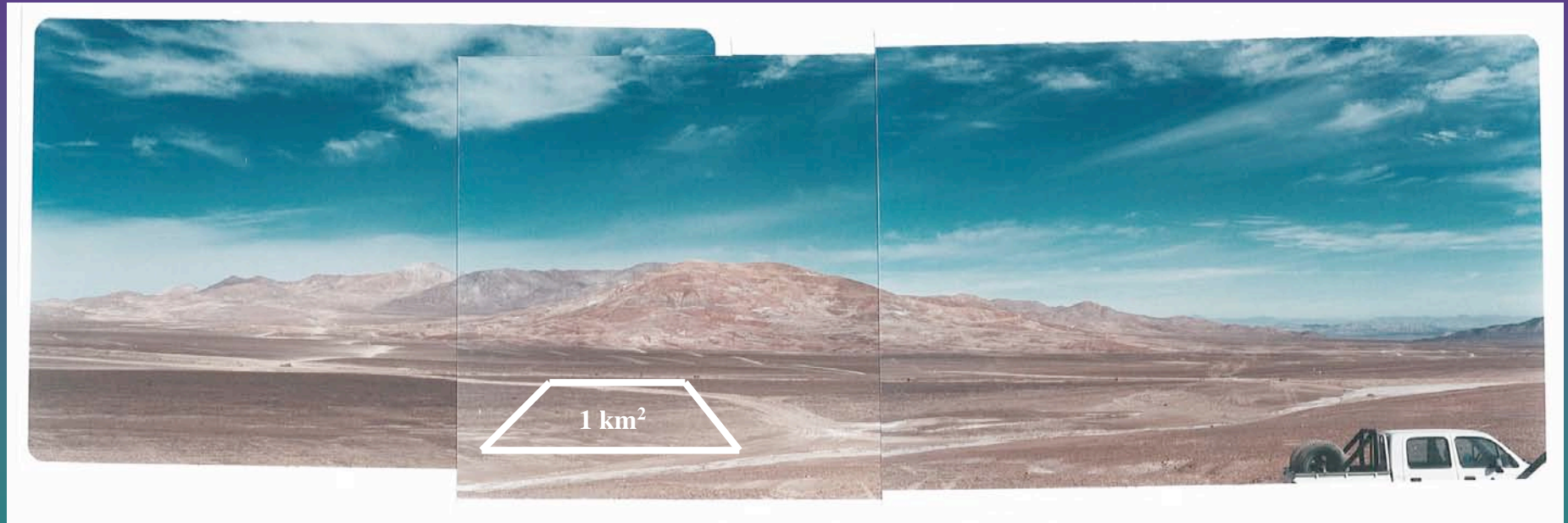
Andradite garnet – $\text{Ca}_3\text{Fe}_2(\text{SiO}_4)_3$



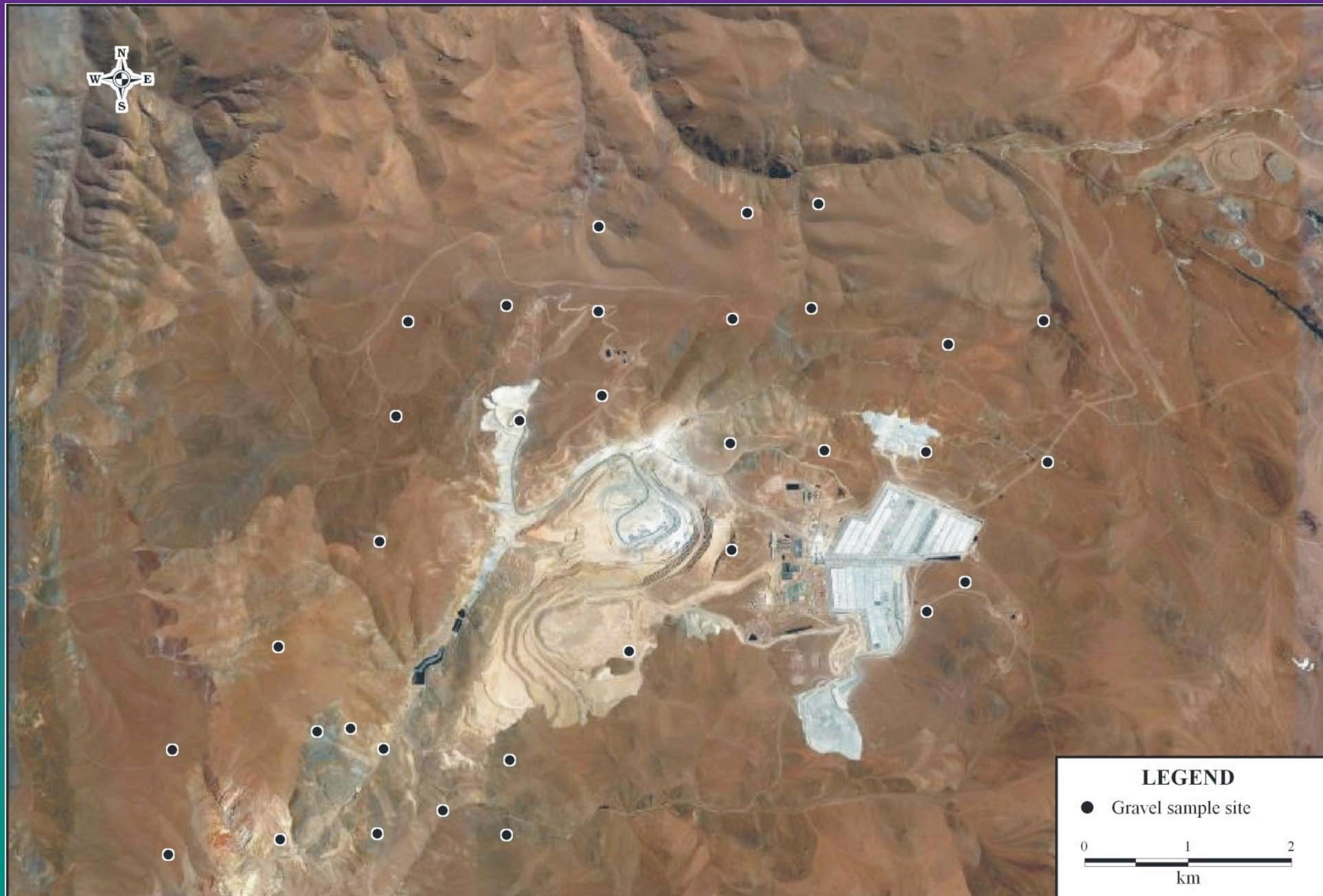
Alteration zones, Escondida, Chile



Arid landscape, Atacama Desert, Chile



Sample sites, Quebrada Blanca

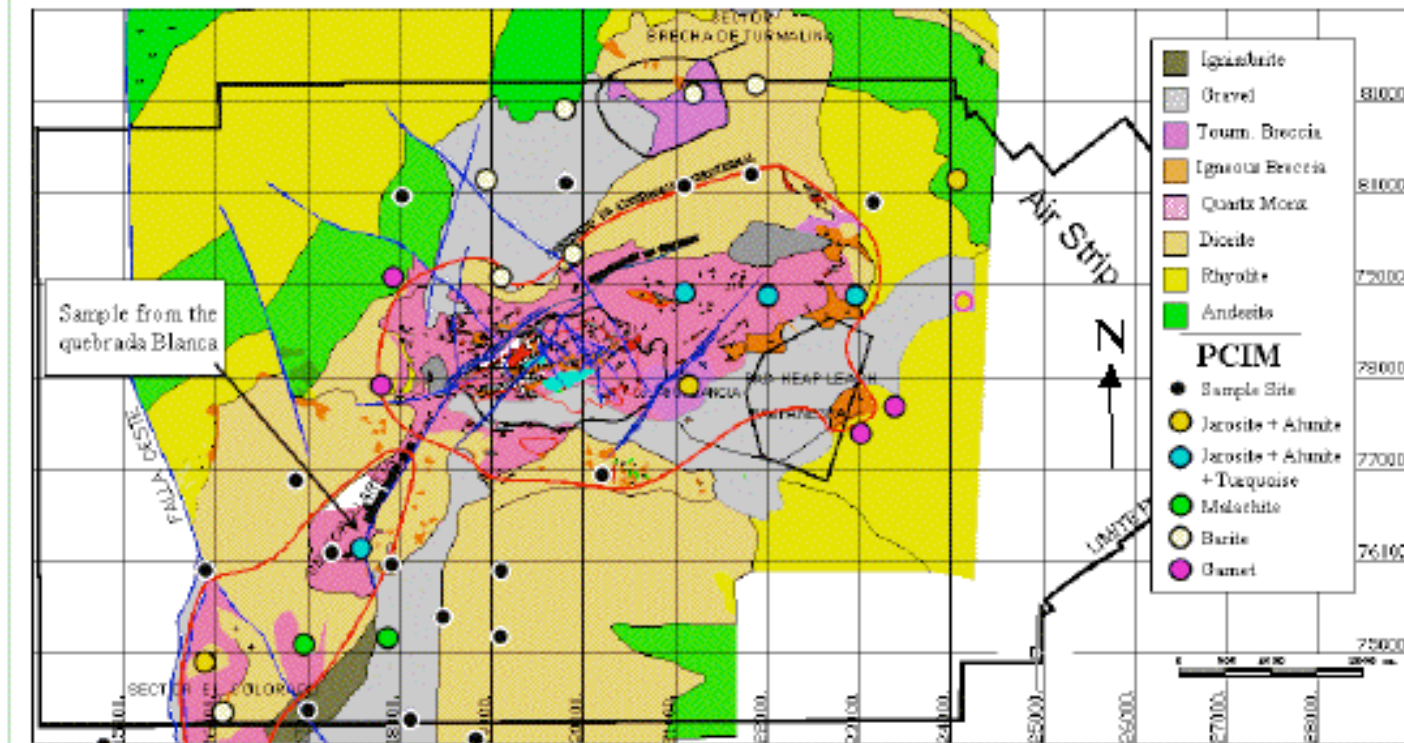


Courtesy: Aur Resources

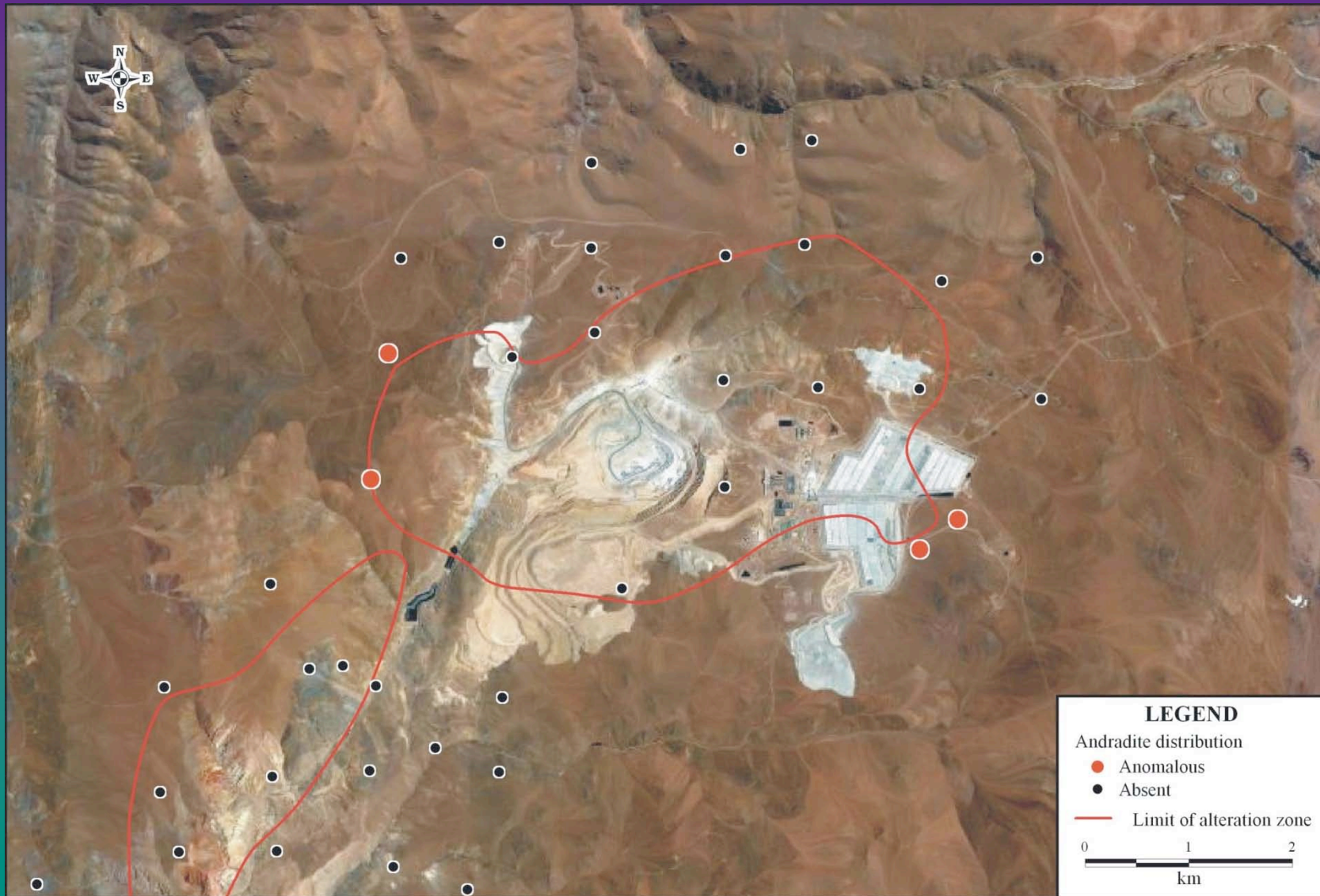


GEOLOGIA DE SUPERFICIE

QUEBRADA BLANCA



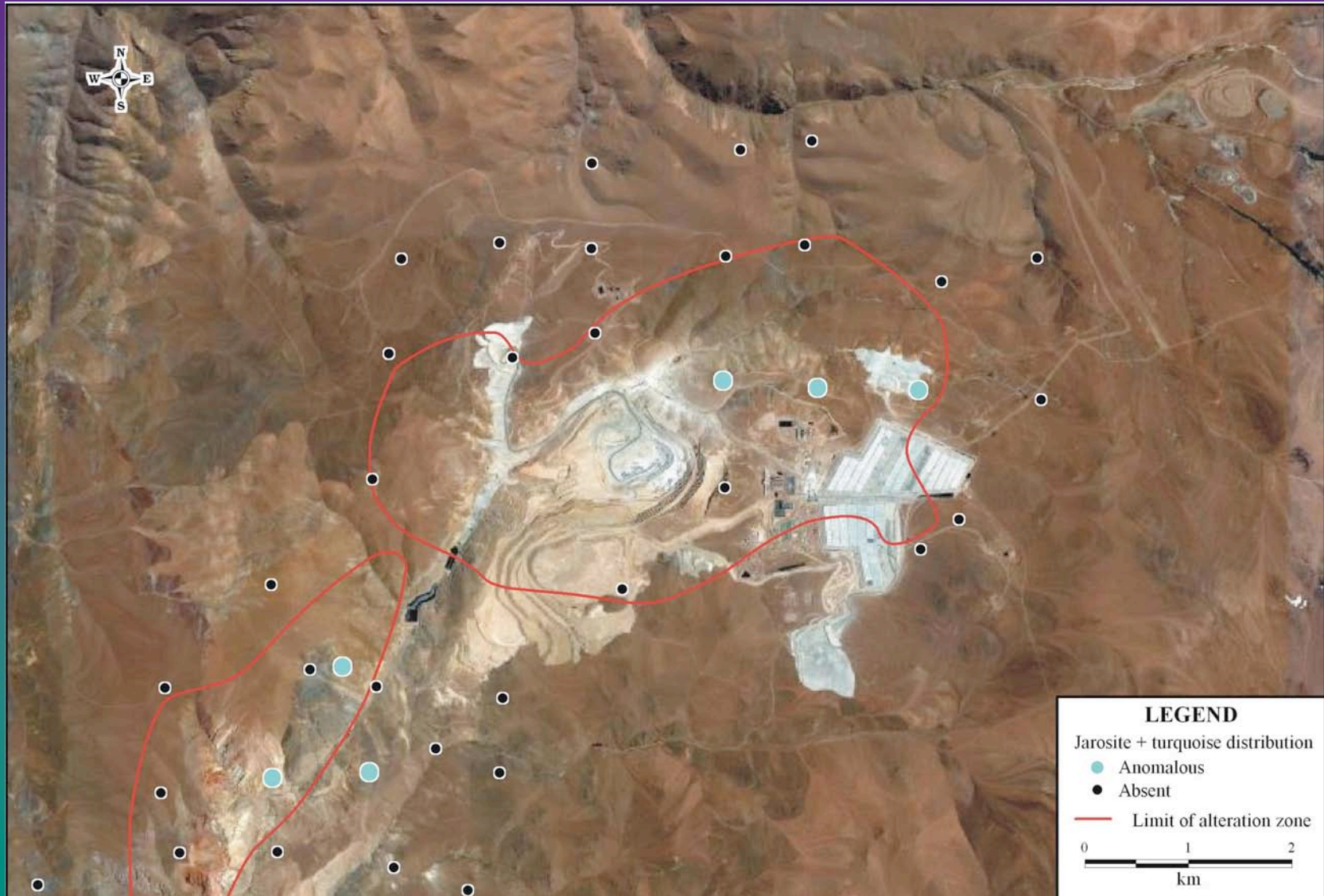
Andradite in alluvium, Quebrada Blanca



Courtesy: Aur Resources



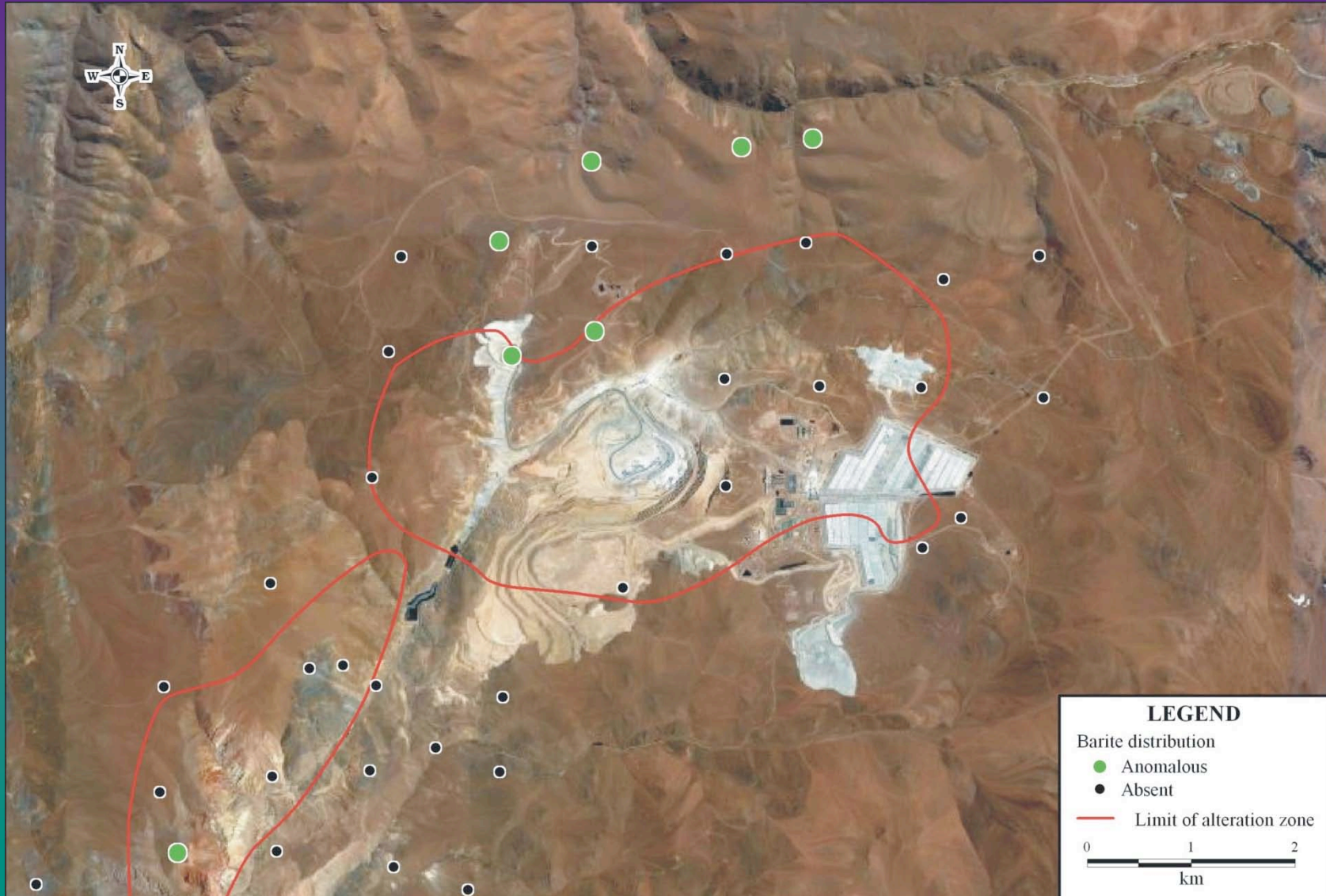
Jarosite + turquoise in alluvium, Quebrada Blanca



Courtesy: Aur Resources



Barite in alluvium, Quebrada Blanca



Courtesy: Aur Resources



Conclusions

Ni-Cu-PGE and porphyry Cu *systems* are very different but they share two key features that are reflected in their indicator mineral footprints:

- Both are *large*. Therefore both have *regional-scale* indicator mineral footprints that can be detected economically with a wide sample spacing (comparable to the footprint of an entire field of kimberlite pipes)
- Both systems are *zoned* in time and space. Each mineralizing event, process or alteration zone supplies a different subsuite of indicator minerals to the regional anomaly. If we *tighten* our sample spacing at the head of this anomaly, we can resolve this zoning and focus on the best targets (comparable to locating the most fertile pipes within a kimberlite field)

Together these features make indicator mineralogy a very effective exploration tool for both Ni-Cu-PGE and porphyry Cu deposits and possibly for other large-scale magmatic-hydrothermal systems such as IOCG.