Geochemical and isotopic characterization of kimberlitic waters a proposal of a new diamond exploration technique

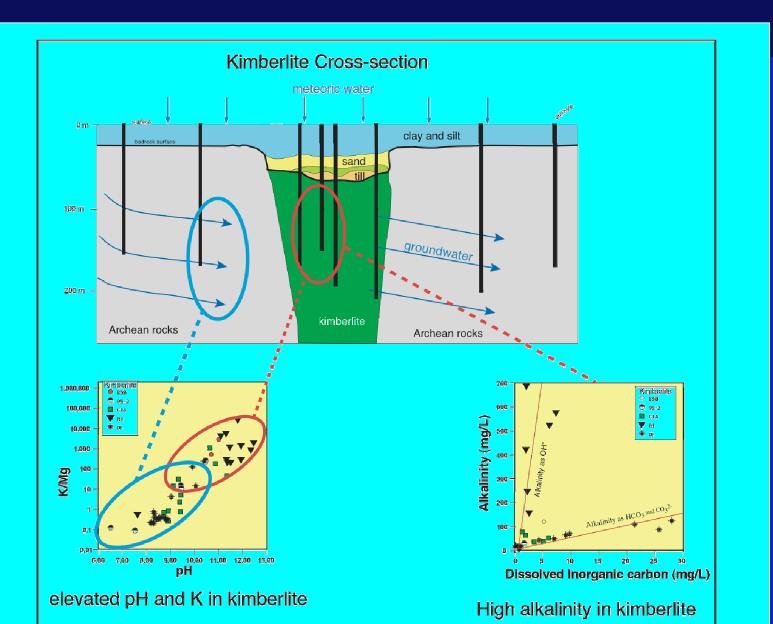
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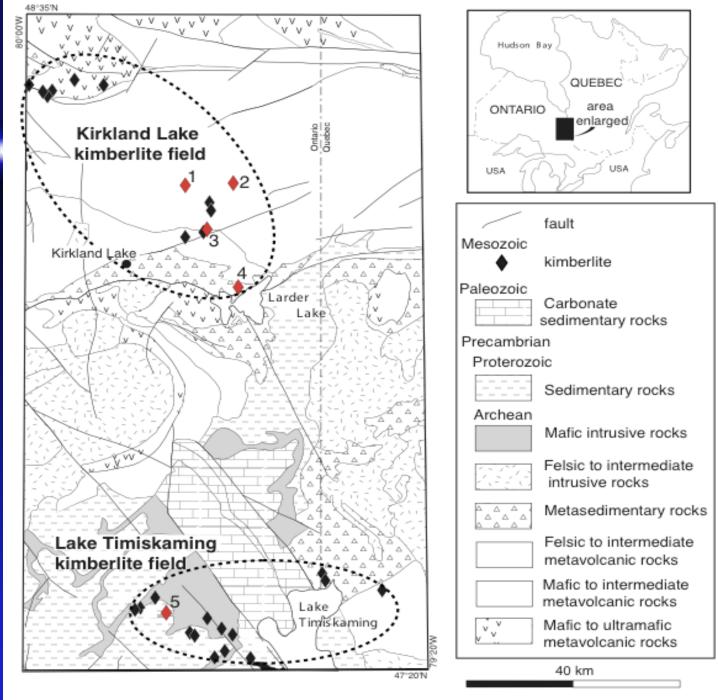






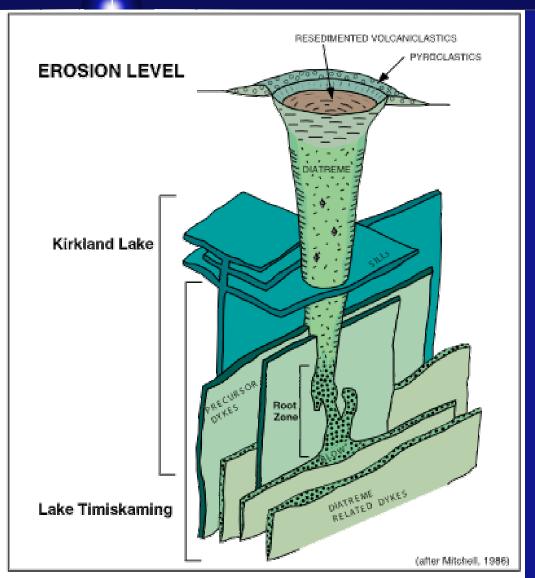
Flow model - Exploration





. Bedrock geology from Ontario Geological Survey (1991).

Characteristics of a Kimberlite



- Kimberlites are ultramafic bodies.
- Kimberlite groundmass is consists mainly of alteration minerals.
- The host rock of kimberlites in northeastern Ontario are Archean metasedimentary, and felsic volcanics.

Currently Used Methods of Kimberlite Exploration

- Indicator mineral methods
- Involves tracing the path of past glaciers and their processes using minerals such as Crpyrope, Cr-diopside and Cr-ilmenite to determine kimberlite location.
- Geophysical methods
- Involves the use of airborne magnetic succeptability in order the locate possible kimberlite targets.

Sampling

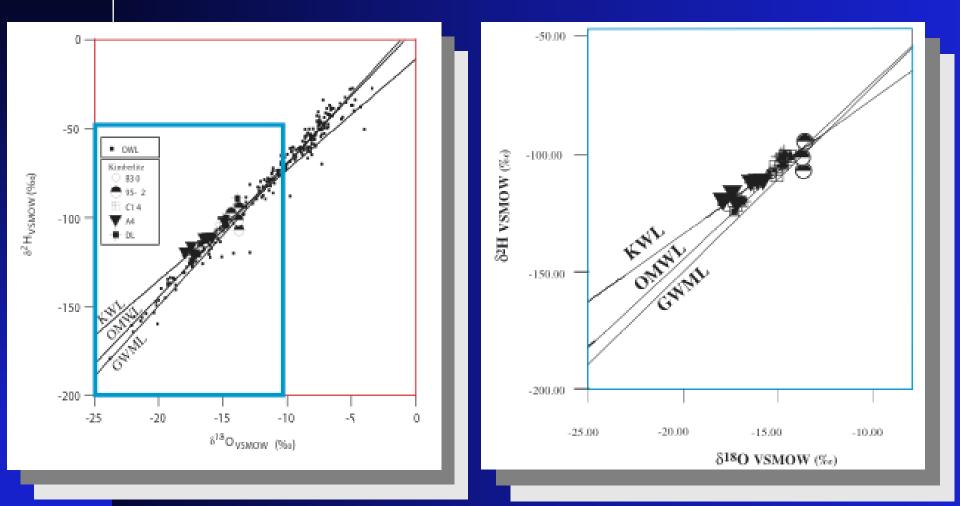


Why use aqeous geochemistry to locate undiscovered kimberlites?

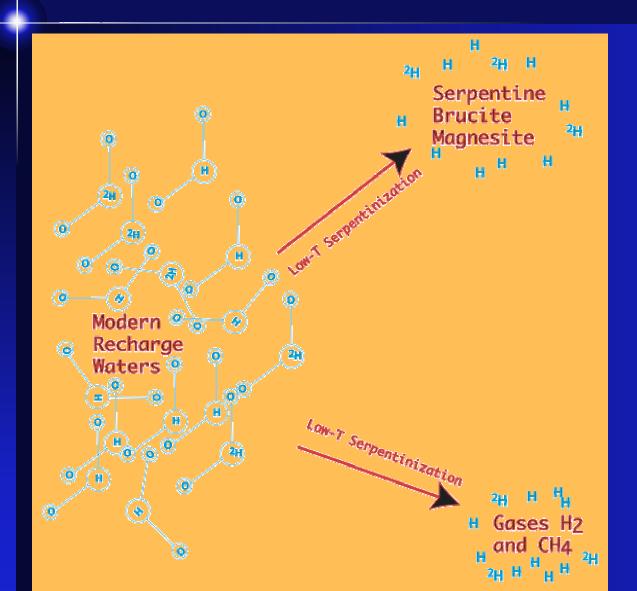
1) Isotopic Characteristics of a Kimberlitic Water

- Deviations from the Local Meteoric Water Line of high pH waters (δ²H and δ¹⁸O ratios)
- Variations in $\delta^{13}C$
- Waters demonstrate low ⁸⁷Sr/⁸⁶Sr ratios

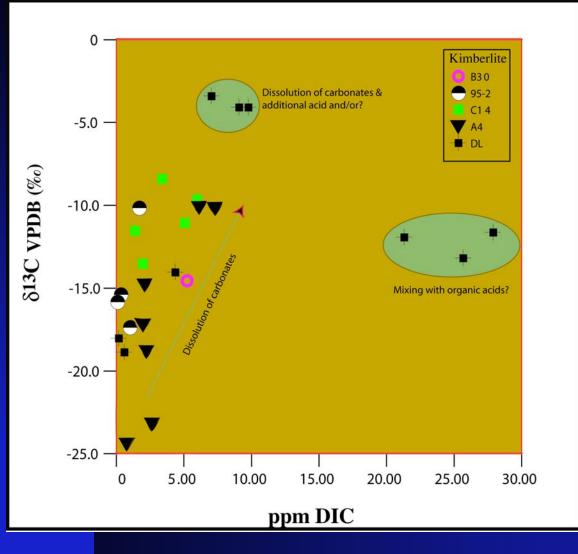
H and O isotopes



Rayleigh Distillation of $\delta^2 H$

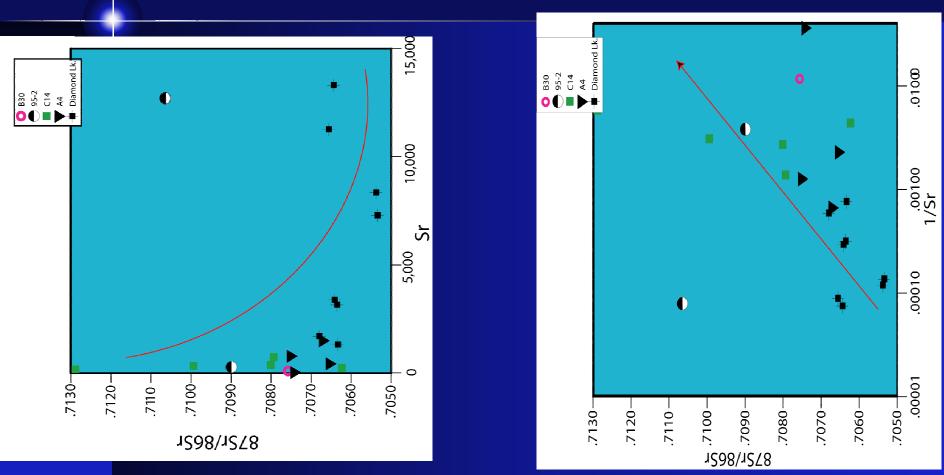


δ^{13} C Variations in waters

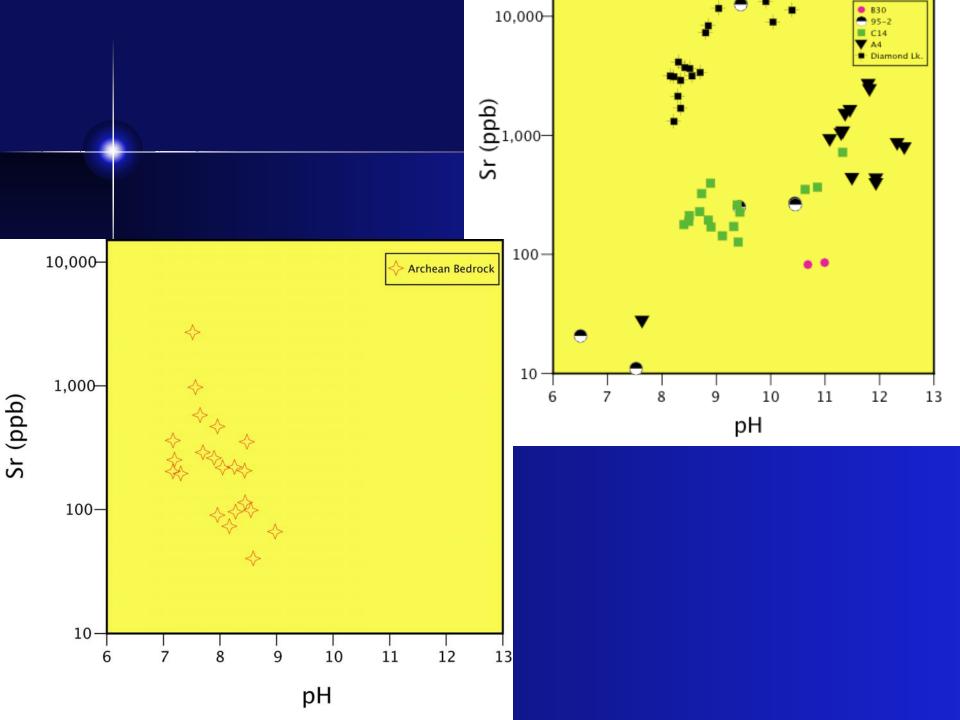


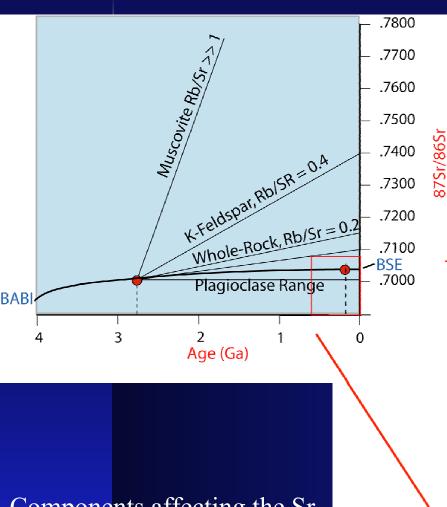
- δ¹³C isotopes in waters demonstrate mixing of C from various sources.
- Paleozoic limestone, mantle carbon and alteration of ratios due to biogenic processes are the main sources.

⁸⁷Sr/⁸⁶Sr isotopic ratios



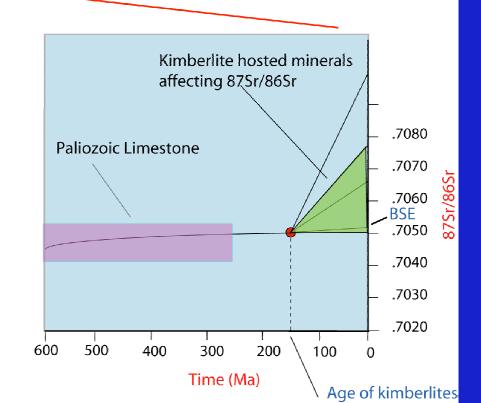
- Kimberlitic waters contain low ⁸⁷Sr/⁸⁶Sr ratios for most samples (average of 0.7065).
- Both graphs indicate that kimberlitic waters are mixing with waters from the Archean host rocks.





Components affecting the Sr ratio in groundwaters in kimberlites. The major components of a granitic rock and the growth of 87Sr/86Sr over time.

Modified from McNutt et al., 1989.





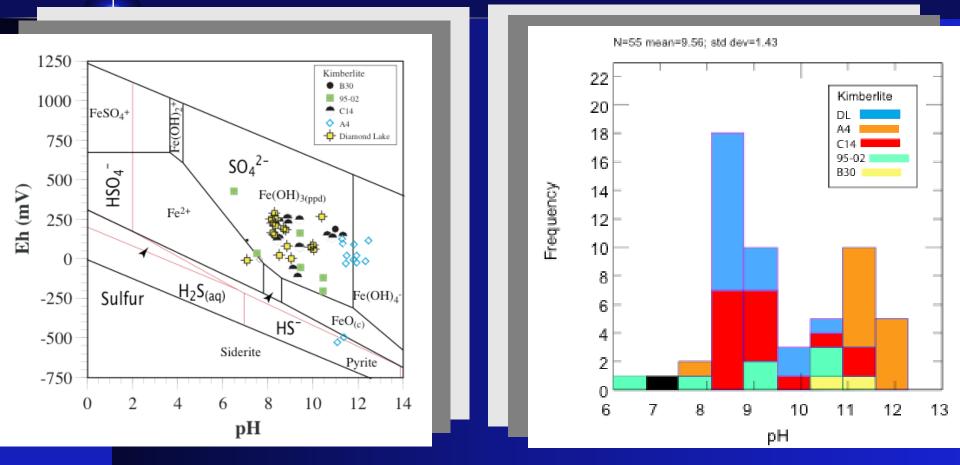




2) Geochemical Characteristics of a Kimberlitic Water

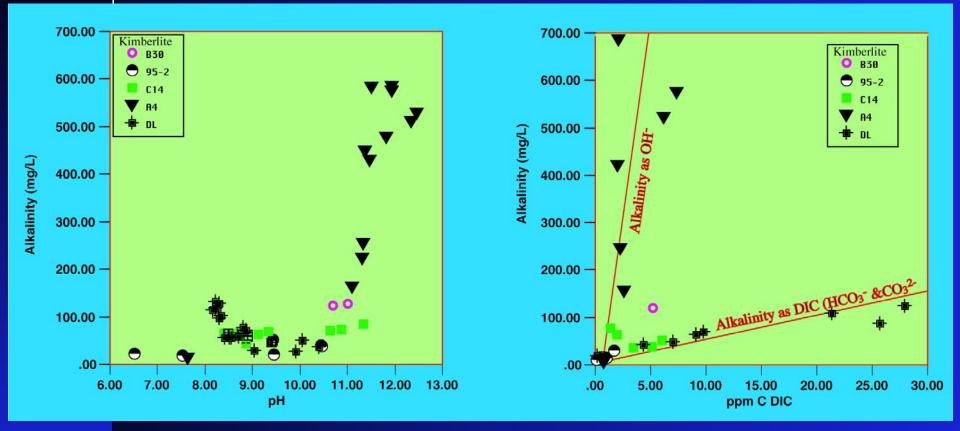
- Strongly reducing waters.
- High alkalinity.
- The relationship that K and the pH have to Mg, Ca and Rb.

Strongly Reducing Waters



 High pH and low Eh in kimberlitic waters contribute to unusual geochemical characteristics in this area of the Superior province.

Alkalinity and pH

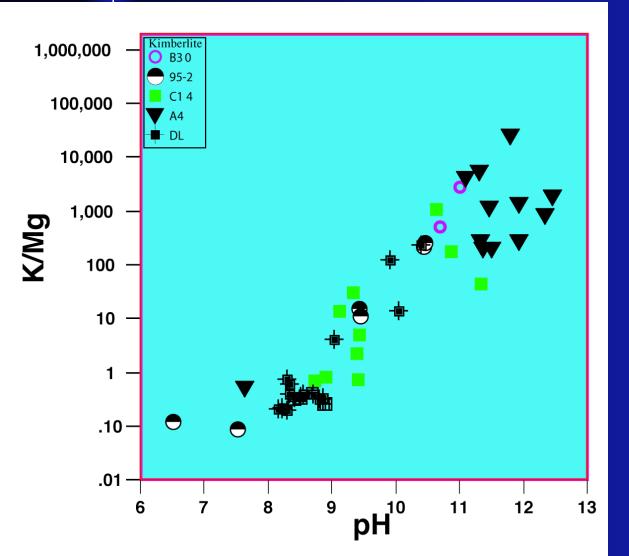


K Concentrations With respect to Mg, Ca and Rb

- Low temperature serpentinization is responsible for the the unusual elemental concentrations in the waters.
- Olivine + H₂O + C (or CO₂) = magnesite + serpentine + CH₄ + brucite + H₂
 (Sherwood Lollar et al., 1993)

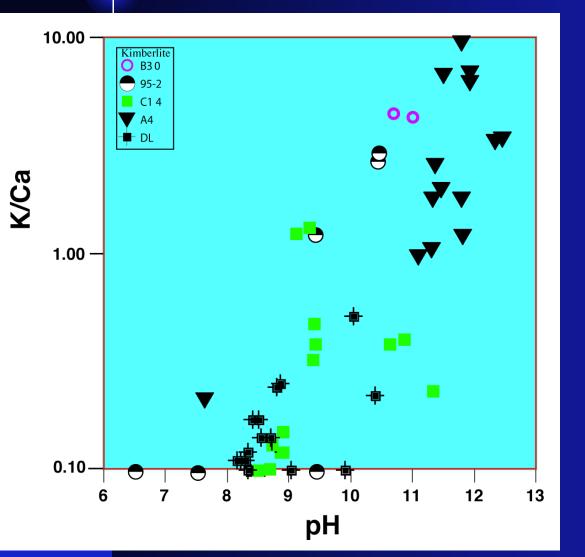
 $5Mg_2SiO_4 + Fe_2SiO_4 + 9H_2O = 3Mg_3Si_2O_5(OH)_4$ (olivine) (serpentine) $+ Mg(OH)_2 + 2Fe(OH)_2$ (brucite) (ferrous hydroxide)

The ratio of K to Mg



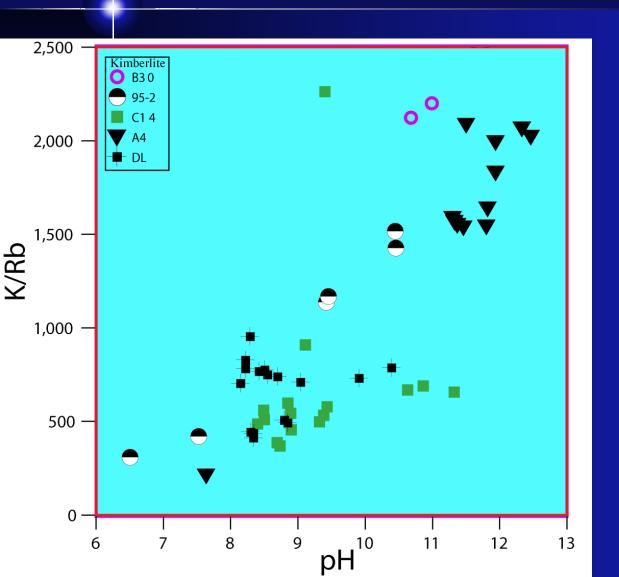
- The ratio of K to Mg increases with pH.
- Mg is becoming increasingly buffered out of the waters when pH becomes high.

The ratio of K to Ca



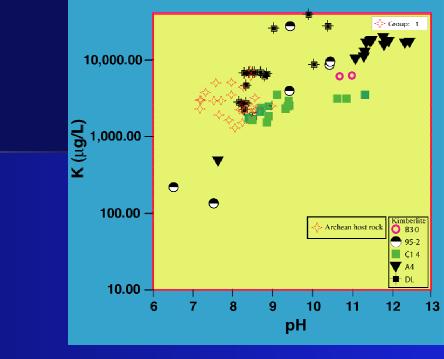
• The Ca is also buffered out of the water with increasing pH when the K remains in the waters.

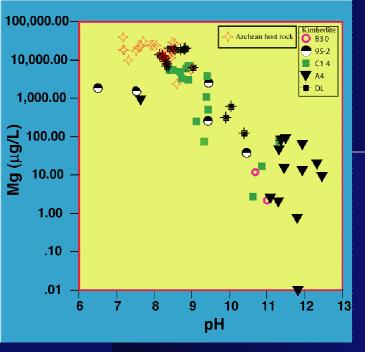
The ratio of K to Rb

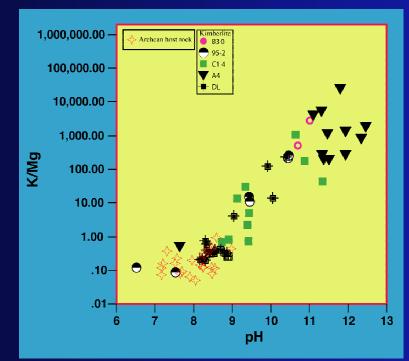


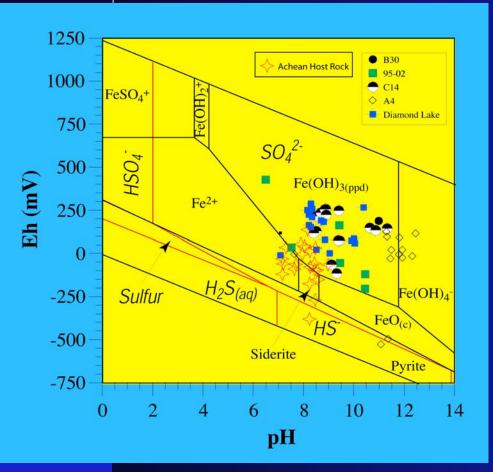
- The Rb is buffered out of the waters compared to K with increasing pH.
- There is question as to why Rb is buffered but not K (ionic radius of the two are almost exact).

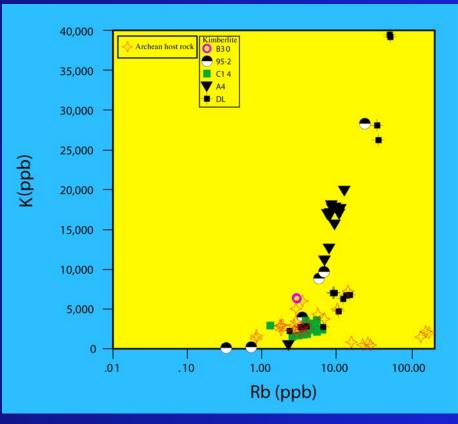
Comparisons with Archean host-rock waters



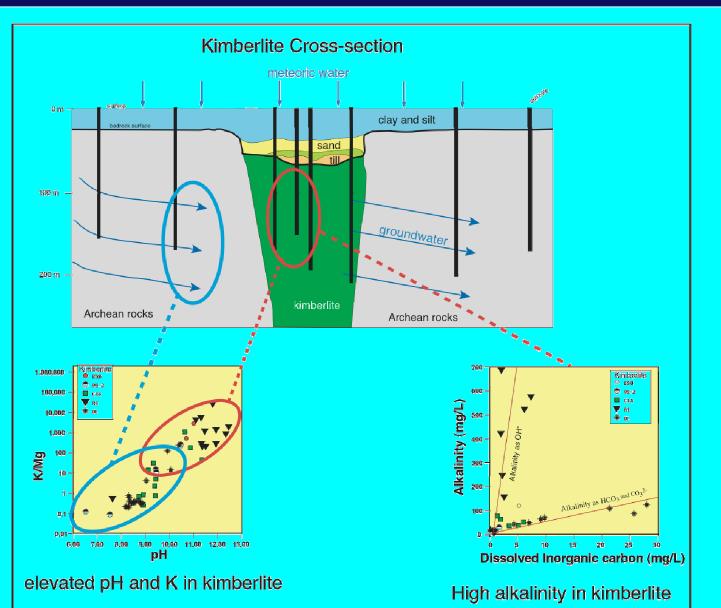








Flow model - Exploration



Future Work

- Complete interpretation of trace element and radiogenic isotope data (can we refine the model to target kimberlites specifically)
- To obtain an accurate mass-balance calculation of the fractionation of $\delta^2 H$ gas samples collected this summer will be analyzed.
- Whole rock and mineral separate geochemical analysis of kimberlite drill core.
- Possible extension of this project to other kimberlite fields (NWT, Arkansas?)

Acknowledgements

- TGI program of the Geological Survey of Canada
- Ontario Geological Survey
- Various mining companies for access to properties and samples of kimberlite

